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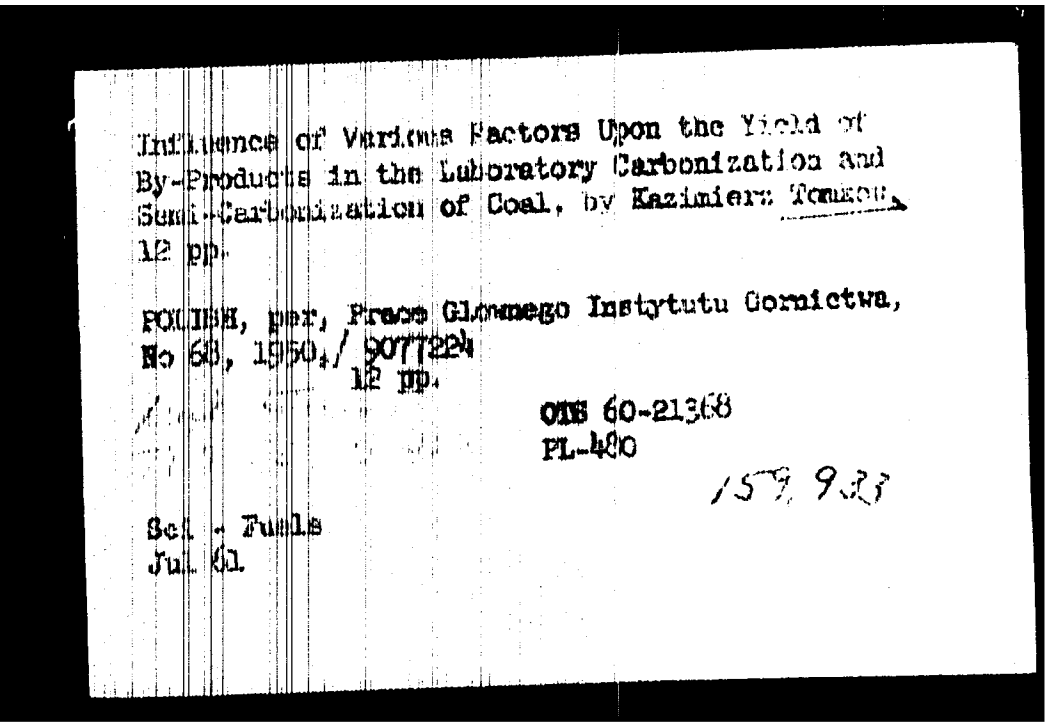

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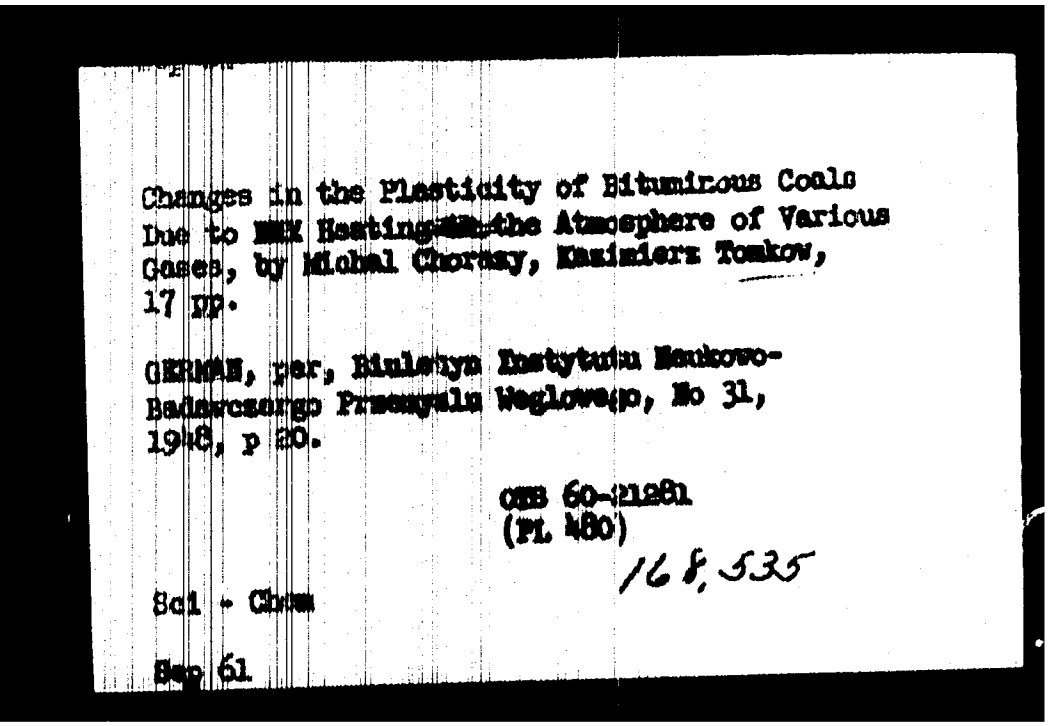

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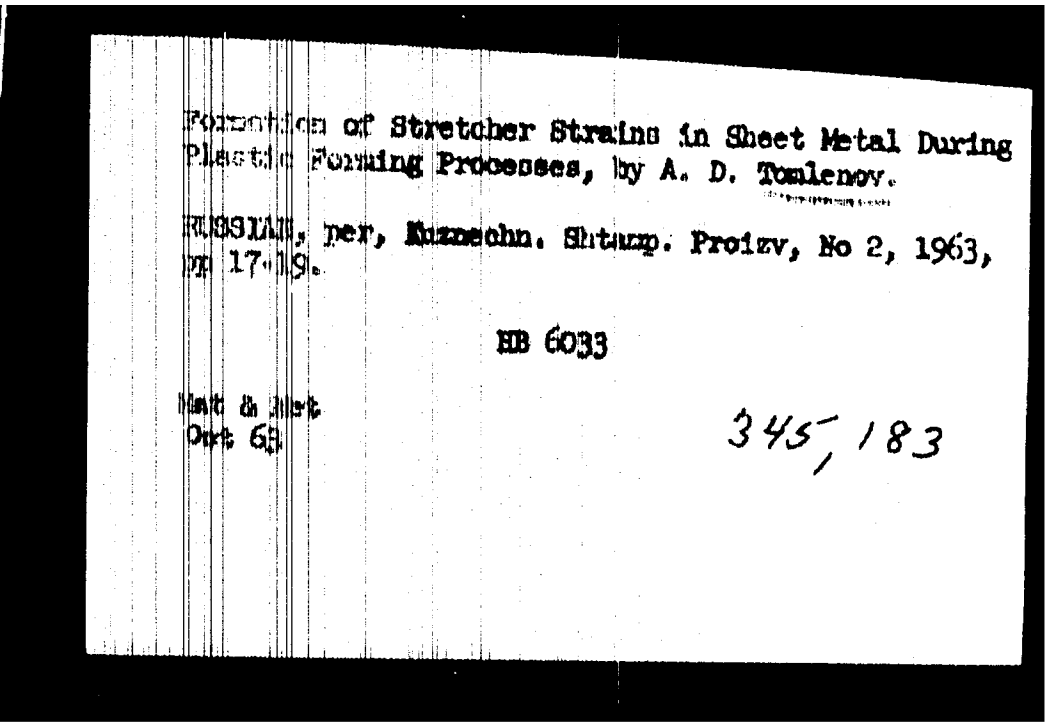

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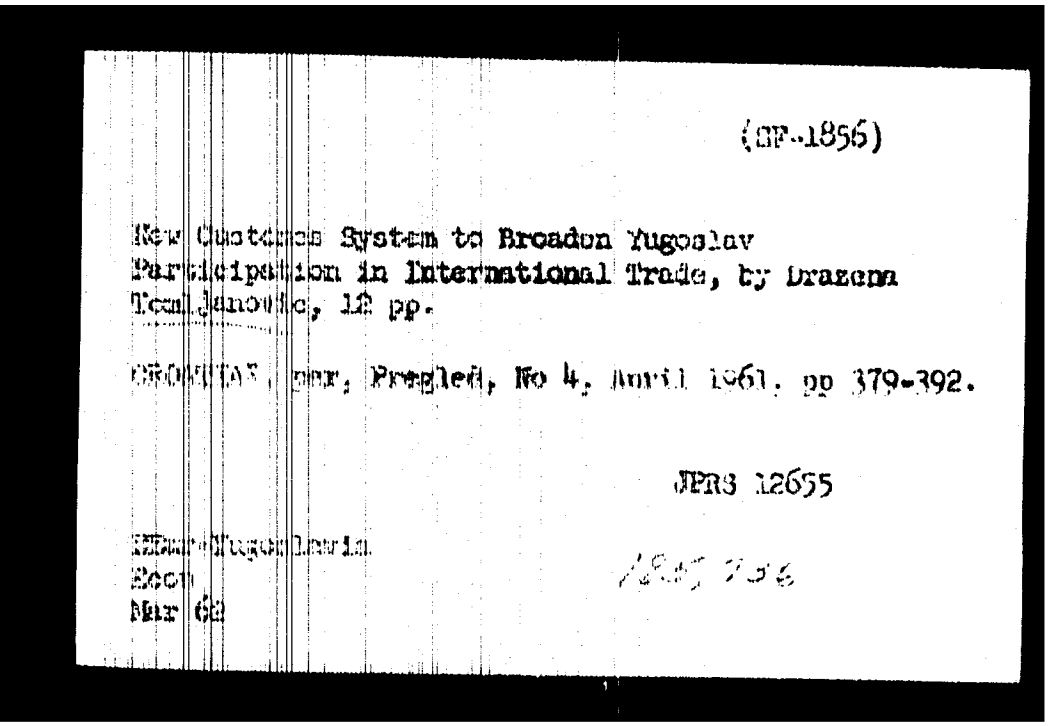

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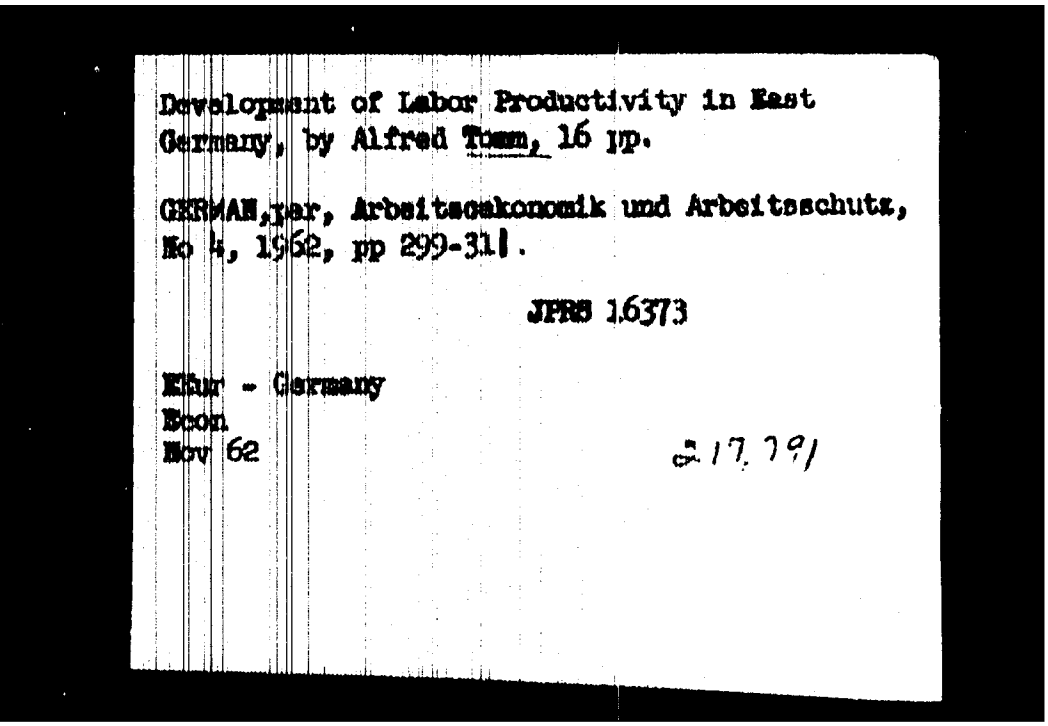

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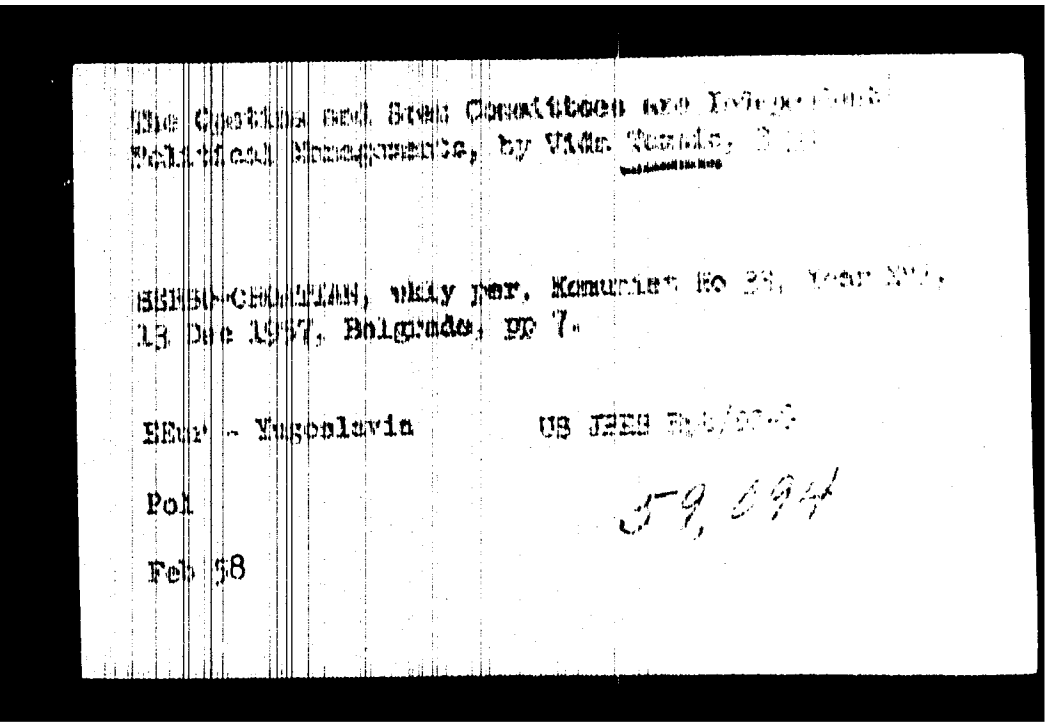

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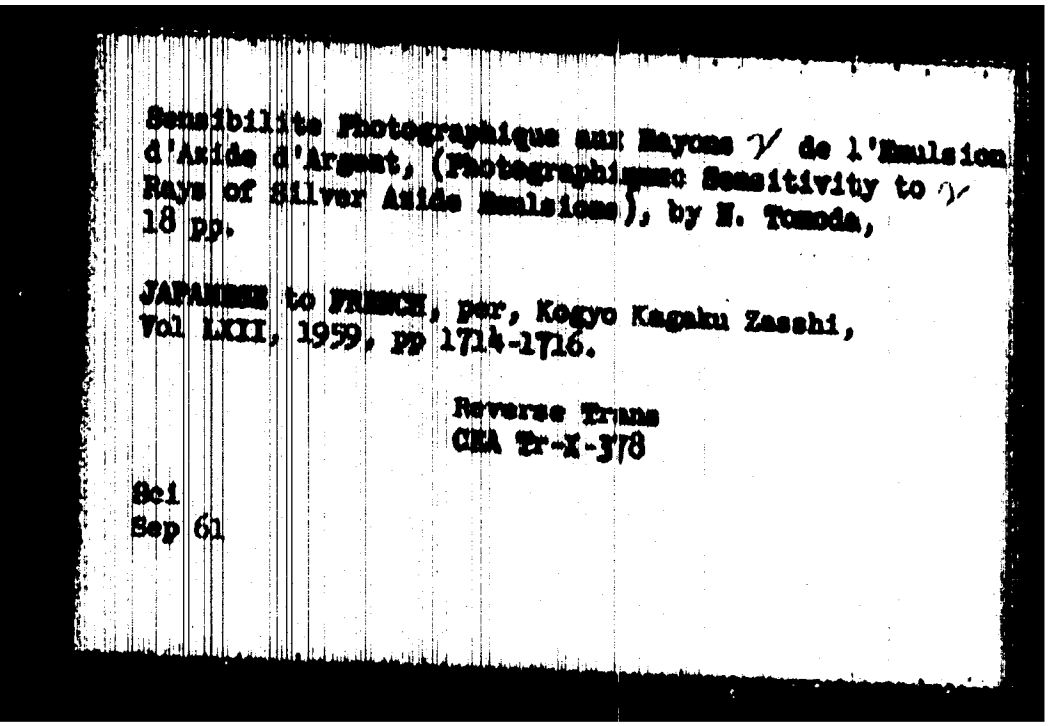

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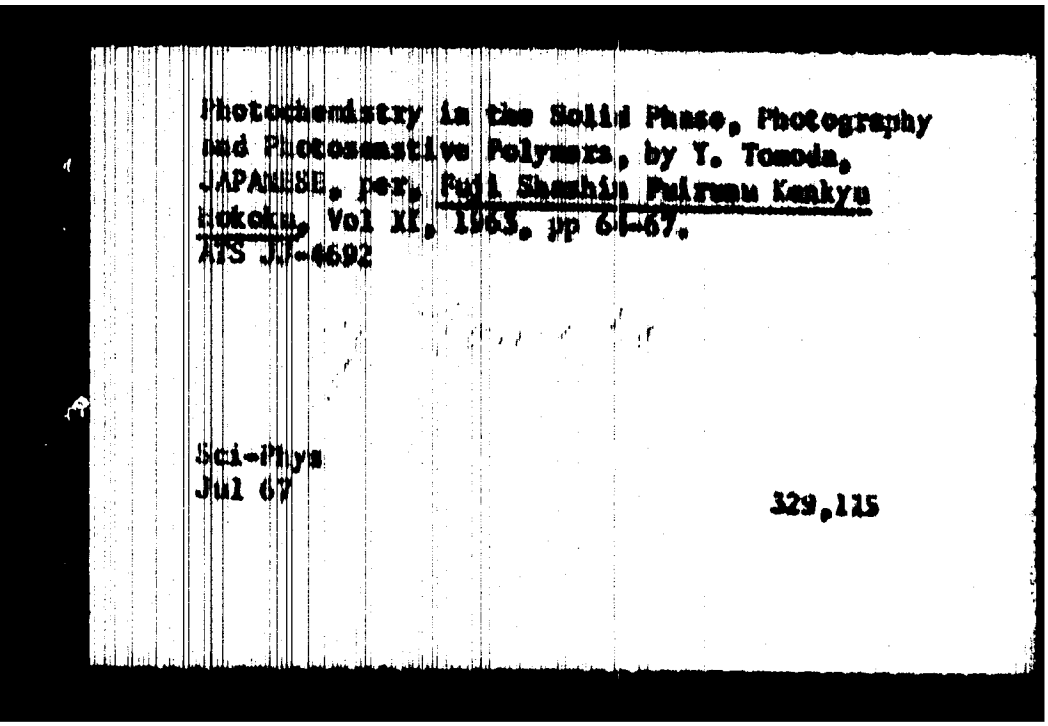

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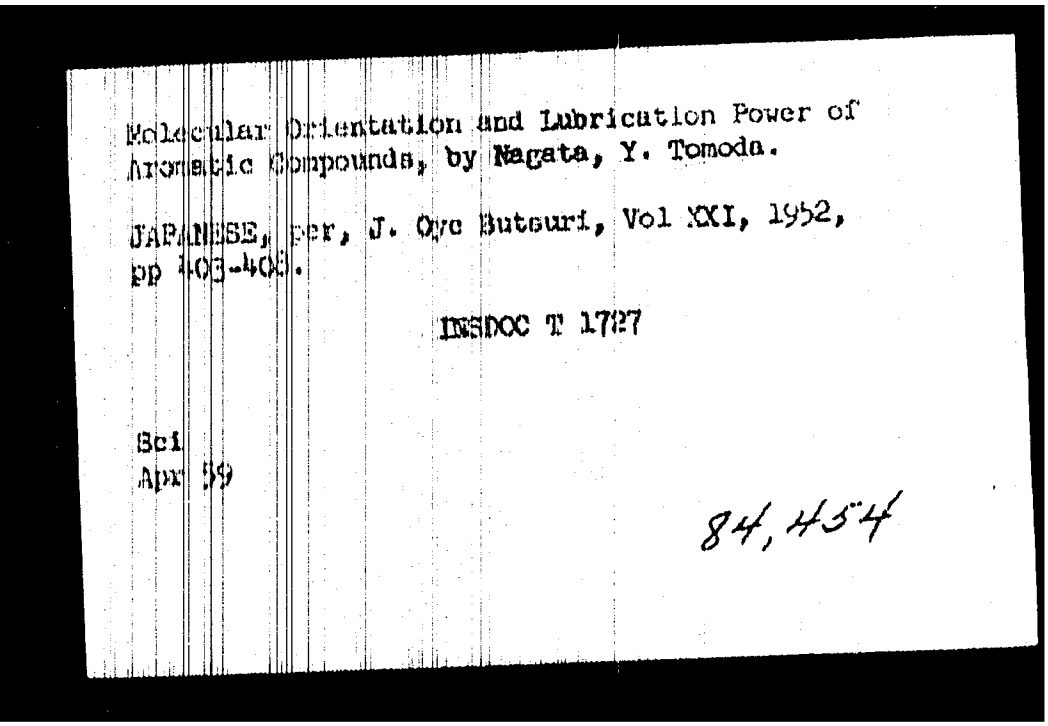

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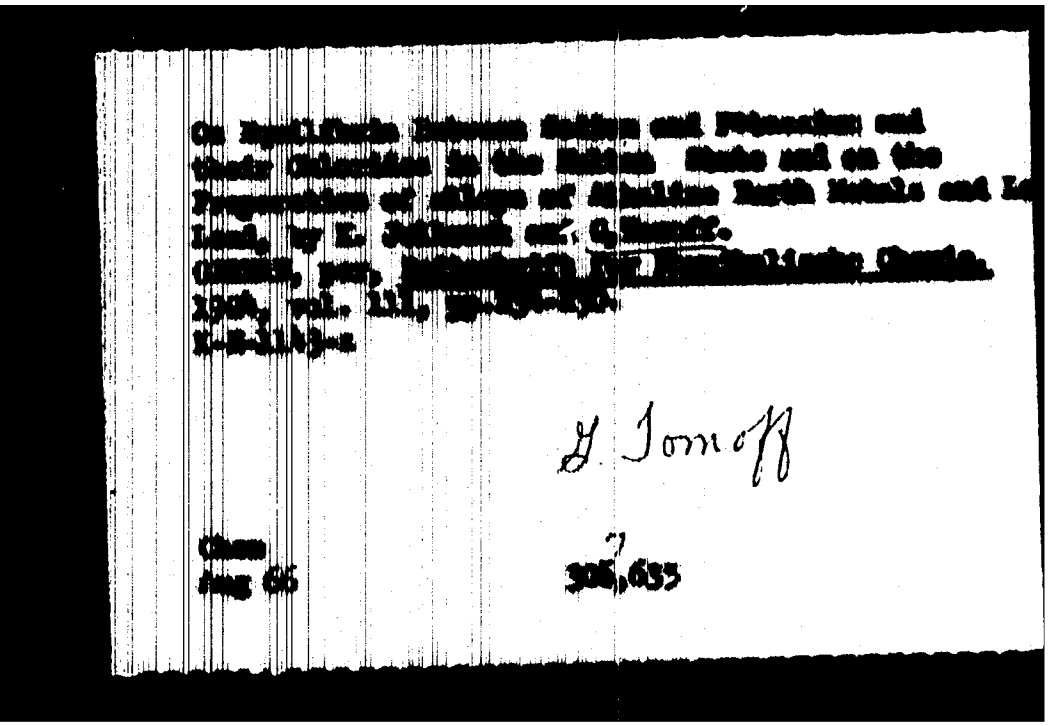

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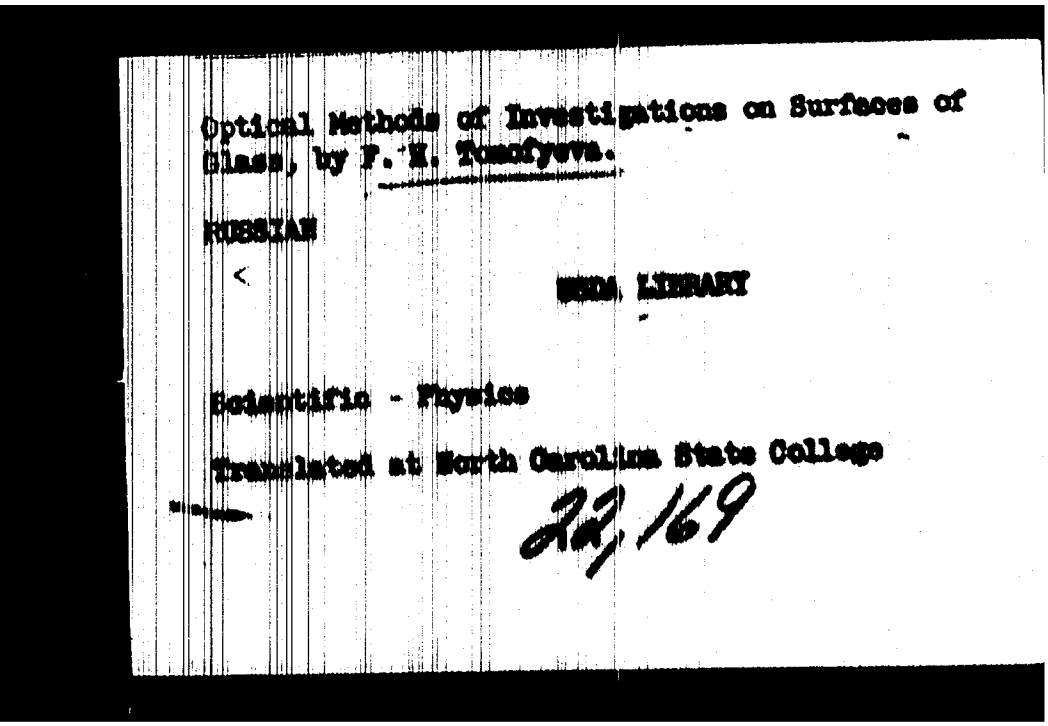

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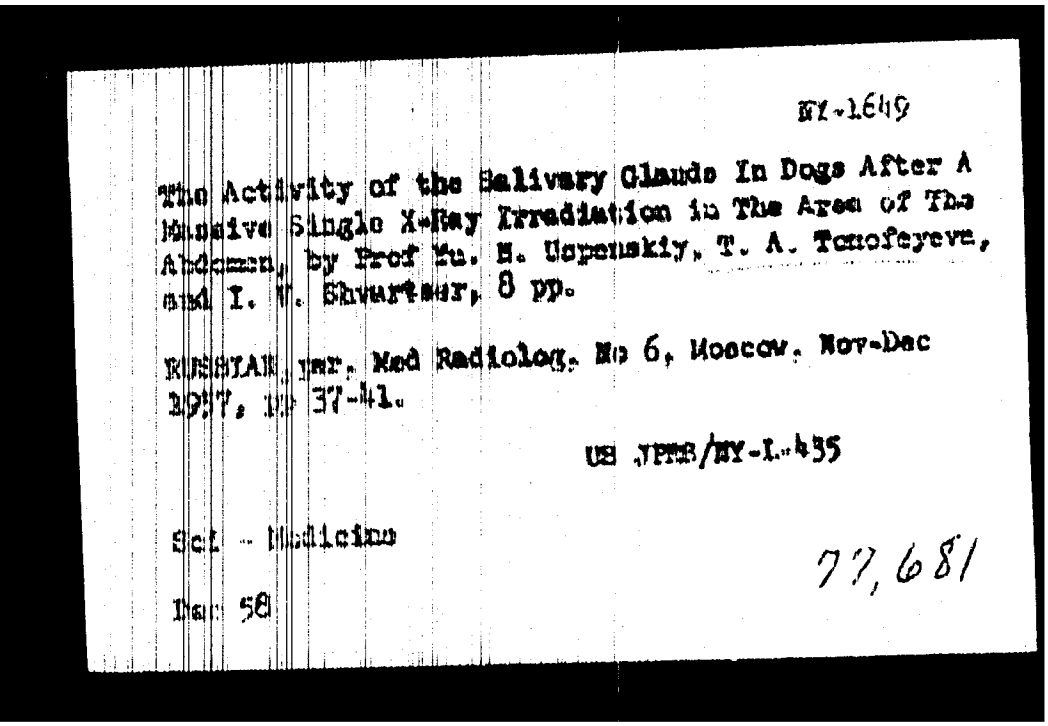

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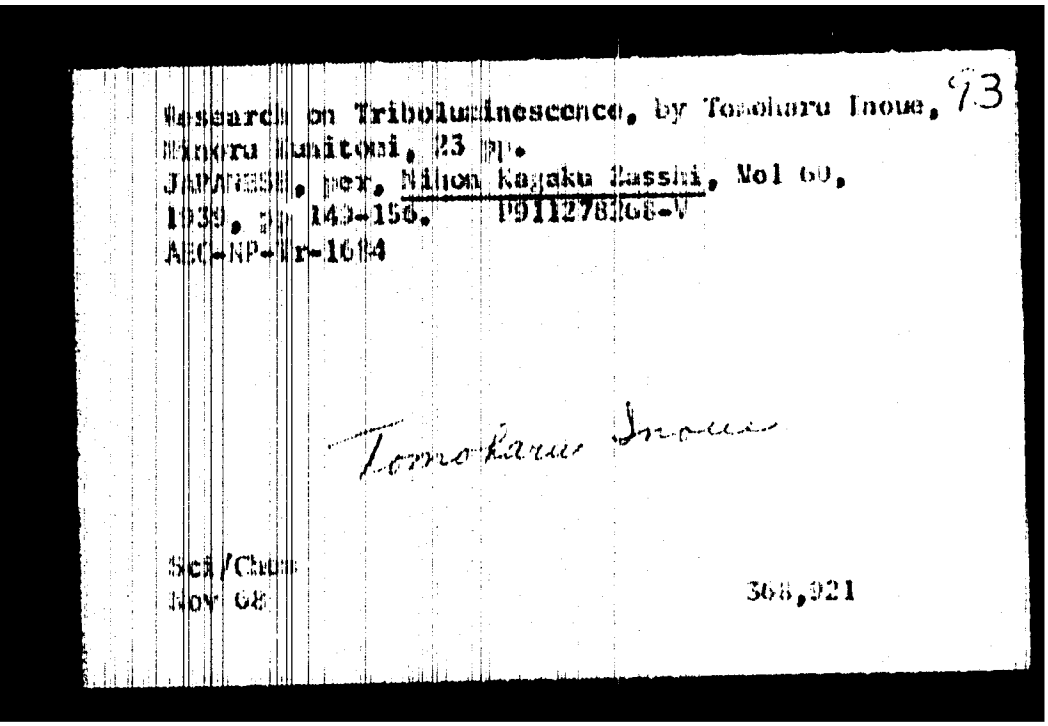

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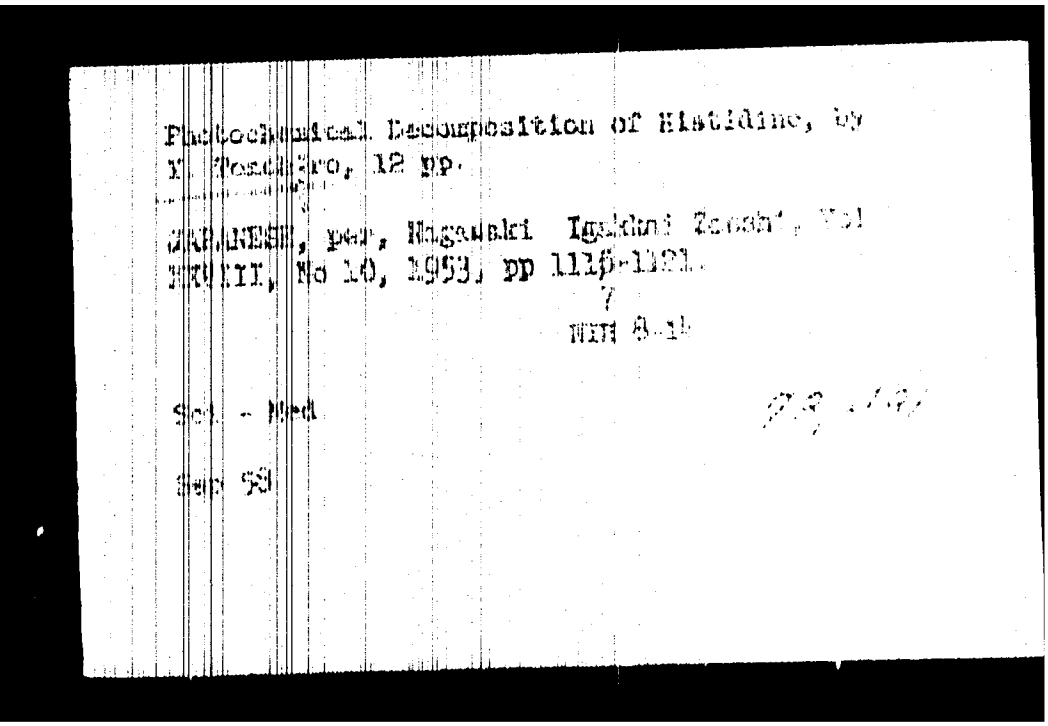

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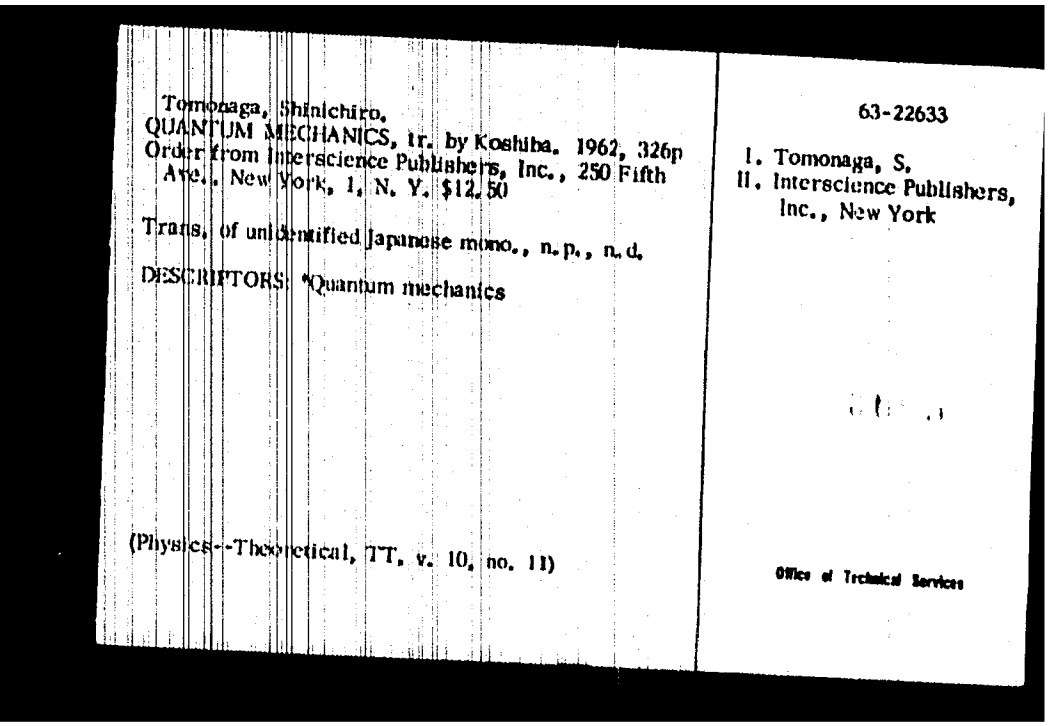

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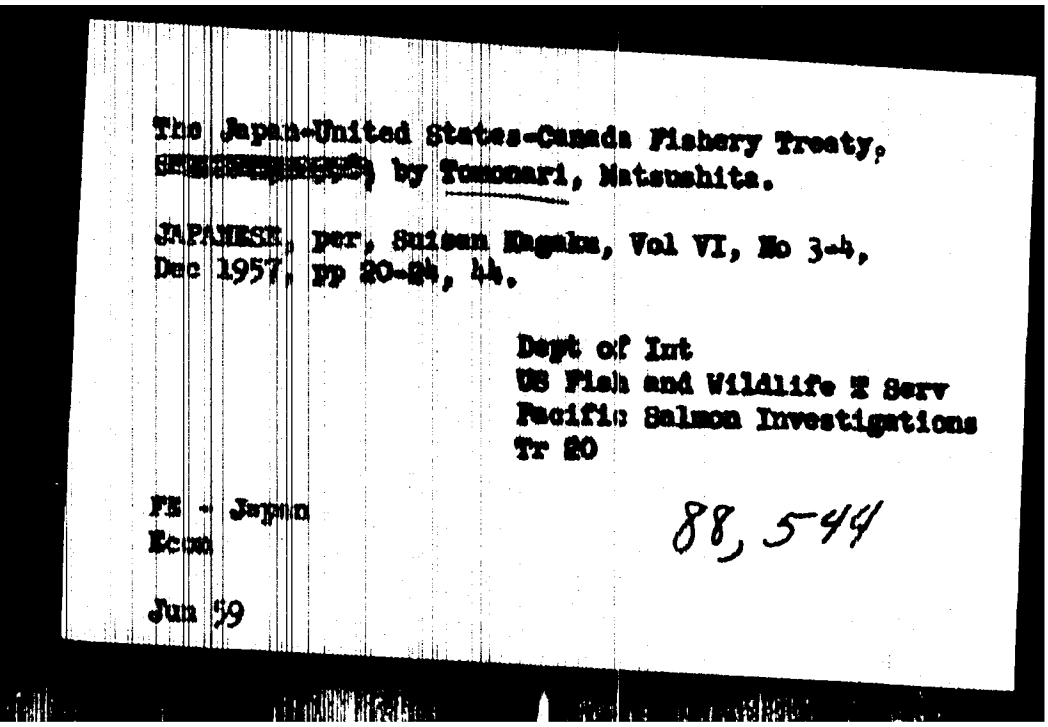

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Order from SLA $1.10 63-18809

I. Tomonari, A.

Trans. of Nihon Kagaku Zasshi (Japan) 1962, v. 83, no. 6, p. 693-695.

DESCRIPTORS: *Chlorides, Ions, Chemical analysis, *Spectroscopy, *Colorimetric analysis, Calibration, *Iron compounds, *Perchlorates, *Mercury compounds, *Thiocyanates, Nitrates.

An iron perchlorate solution and a mercuric thiocyanate solution were added to a specimen solution and the chloride ion was determined by measuring the absorbency of an orange-colored aqueous solution at (Chemistry--Analytical, TT, v. 10, no. 12) (over)

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460 mμ(mμ). It was possible to determine from 0.1 to 15 ppm. of chloride ion speedily and accurately. (Author)

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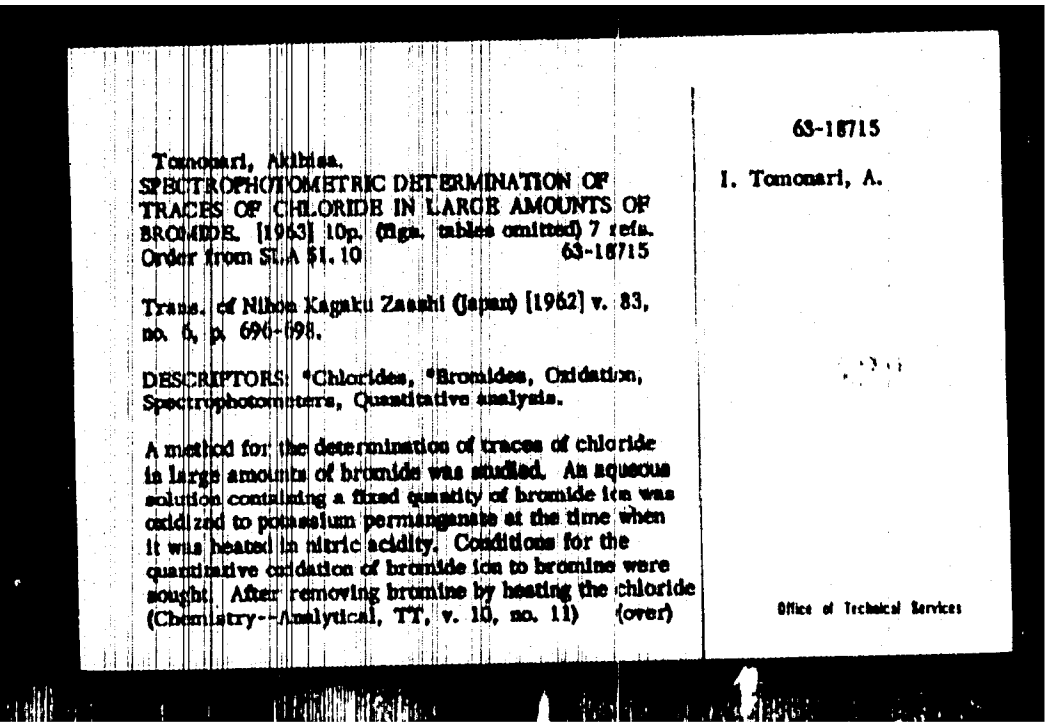

Tomonari, Akihisa.
SPECTROPHOTOMETRIC DETERMINATION OF TRACES OF CHLORIDE IN LARGE AMOUNTS OF BROMIDE. [1963] 10p. (figs. tables omitted) 7 refs. Order from STA $1.10 63-18715

Trans. of Nihon Kagaku Zasshi (Japan) [1962] v. 83, no. 6, p. 696-698.

DESCRIPTORS: *Chlorides, *Bromides, Oxidation, Spectrophotometers, Quantitative analysis.

A method for the determination of traces of chloride in large amounts of bromide was studied. An aqueous solution containing a fixed quantity of bromide ion was oxidized to potassium permanganate at the time when it was heated in nitric acidity. Conditions for the quantitative oxidation of bromide ion to bromine were sought. After removing bromine by heating the chloride (Chemistry--Analytical, TT, v. 10, no. 11) (over)

63-18715

I. Tomonari, A.

Office of Technical Services

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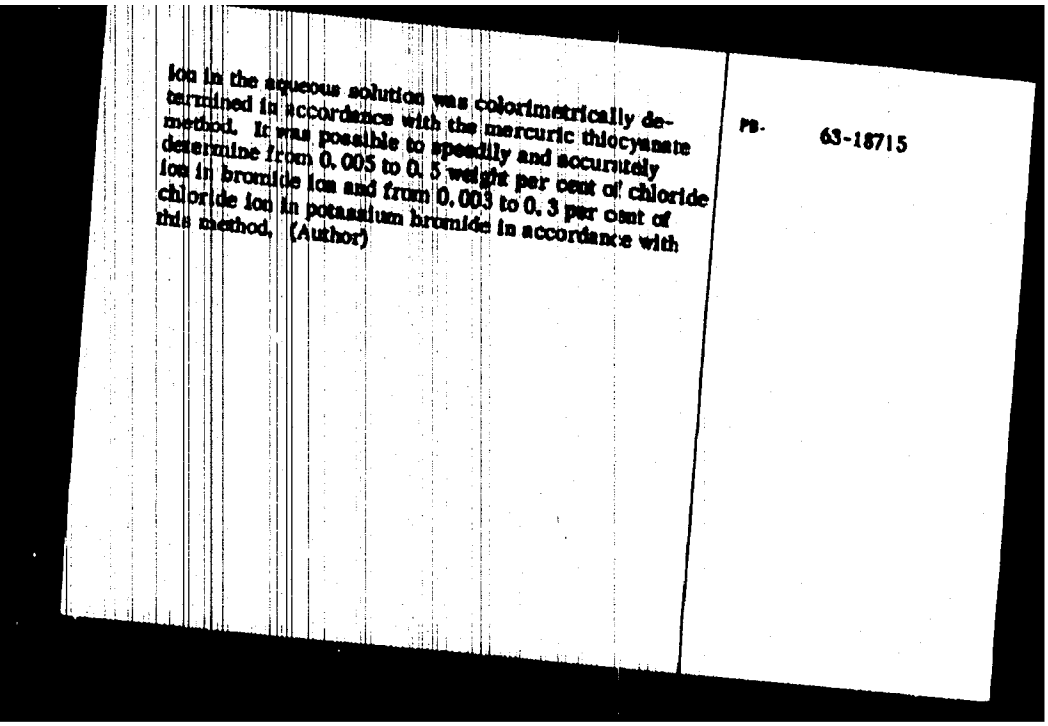

ion in the aqueous solution was colorimetrically determined in accordance with the mercuric thiocyanate method. It was possible to speedily and accurately determine from 0.005 to 0.5 weight per cent of chloride ion in bromide ion and from 0.003 to 0.3 per cent of chloride ion in potassium bromide in accordance with this method. (Author)

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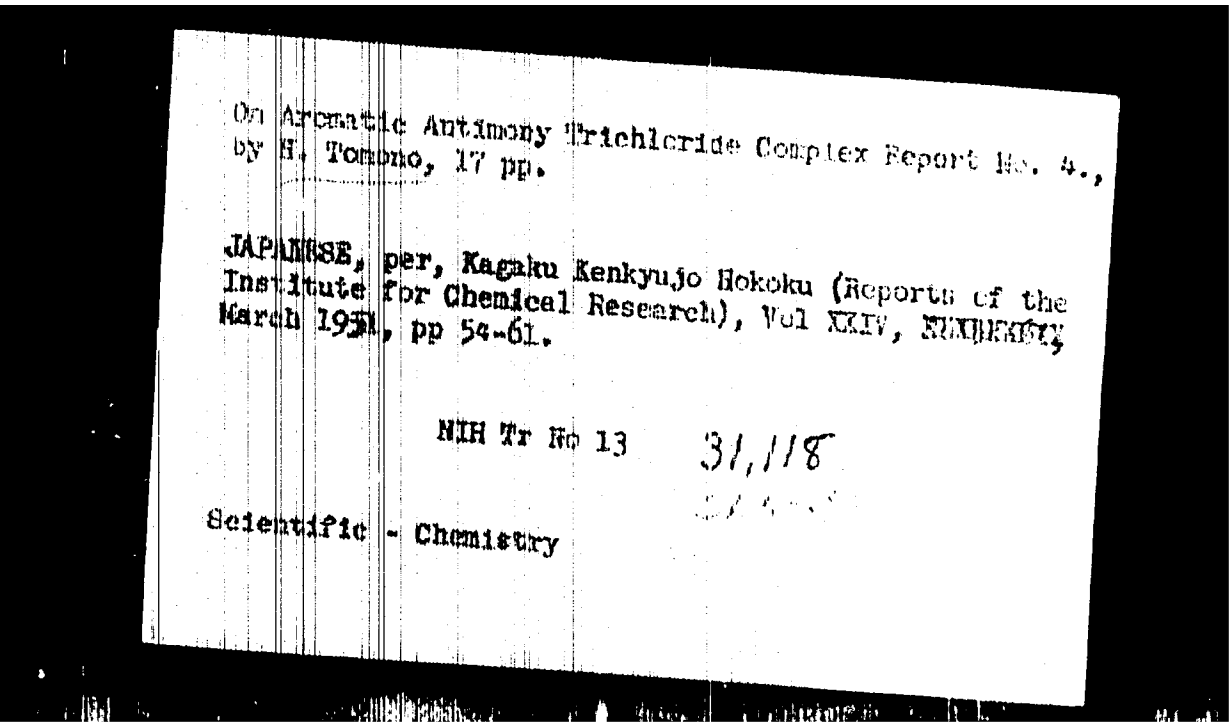

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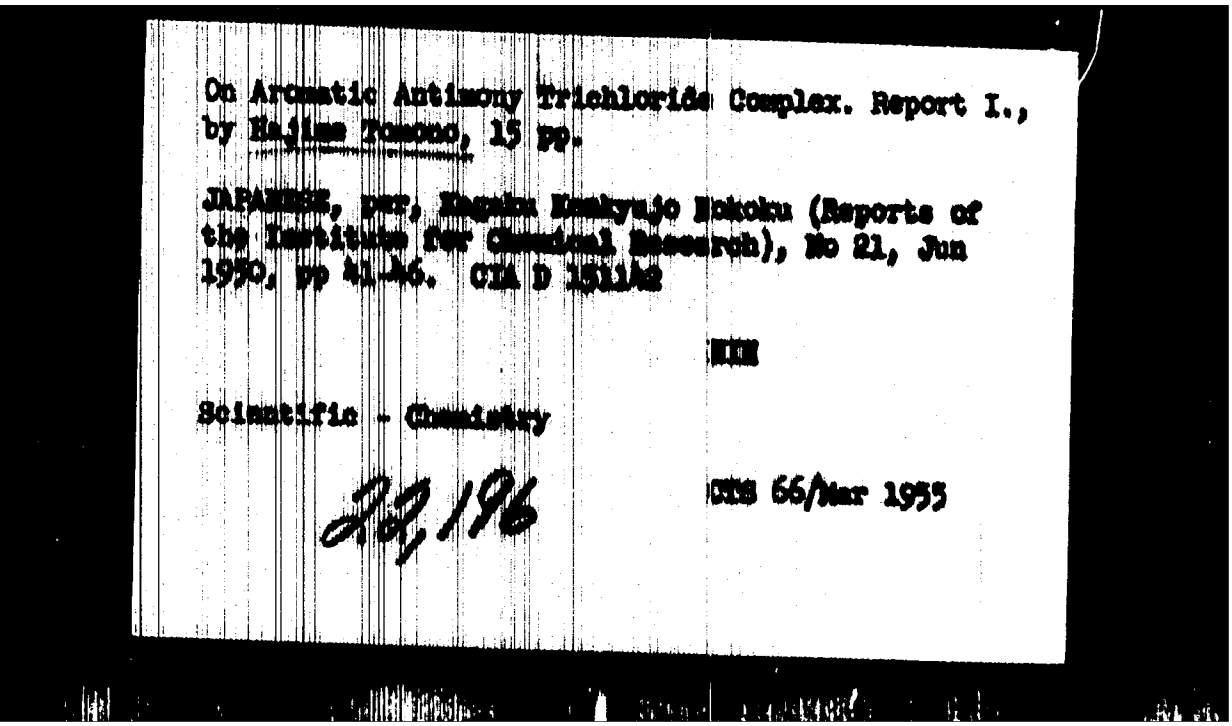

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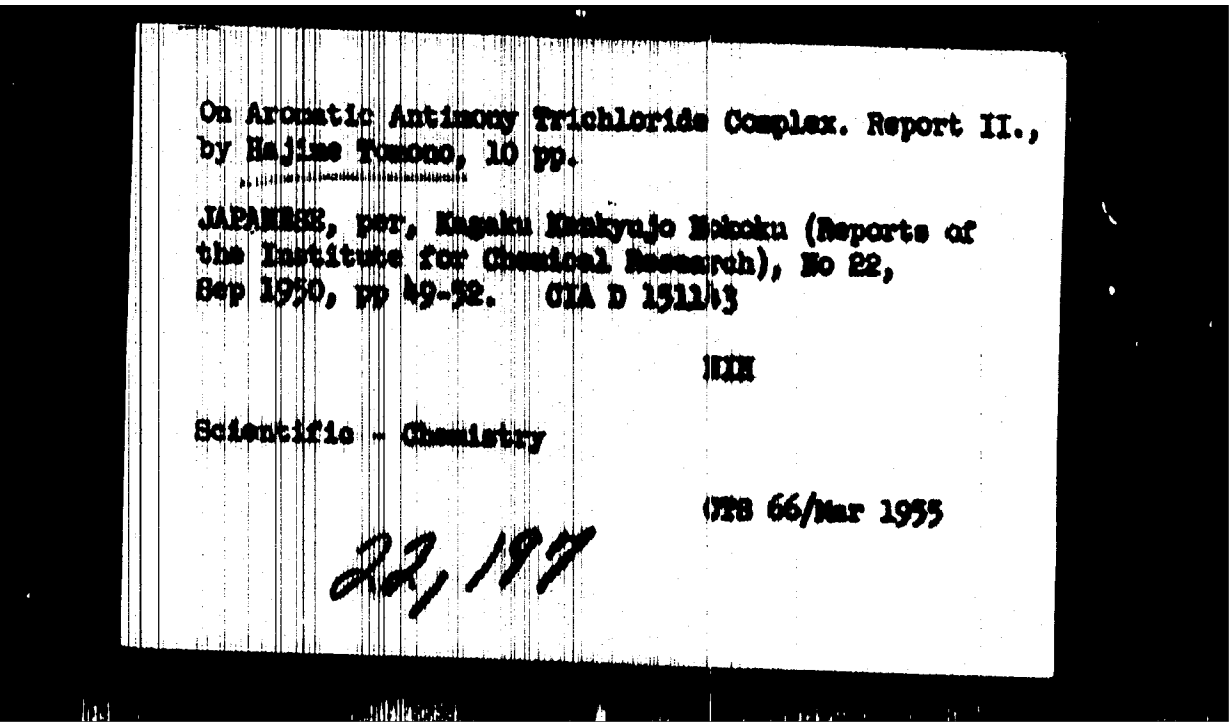

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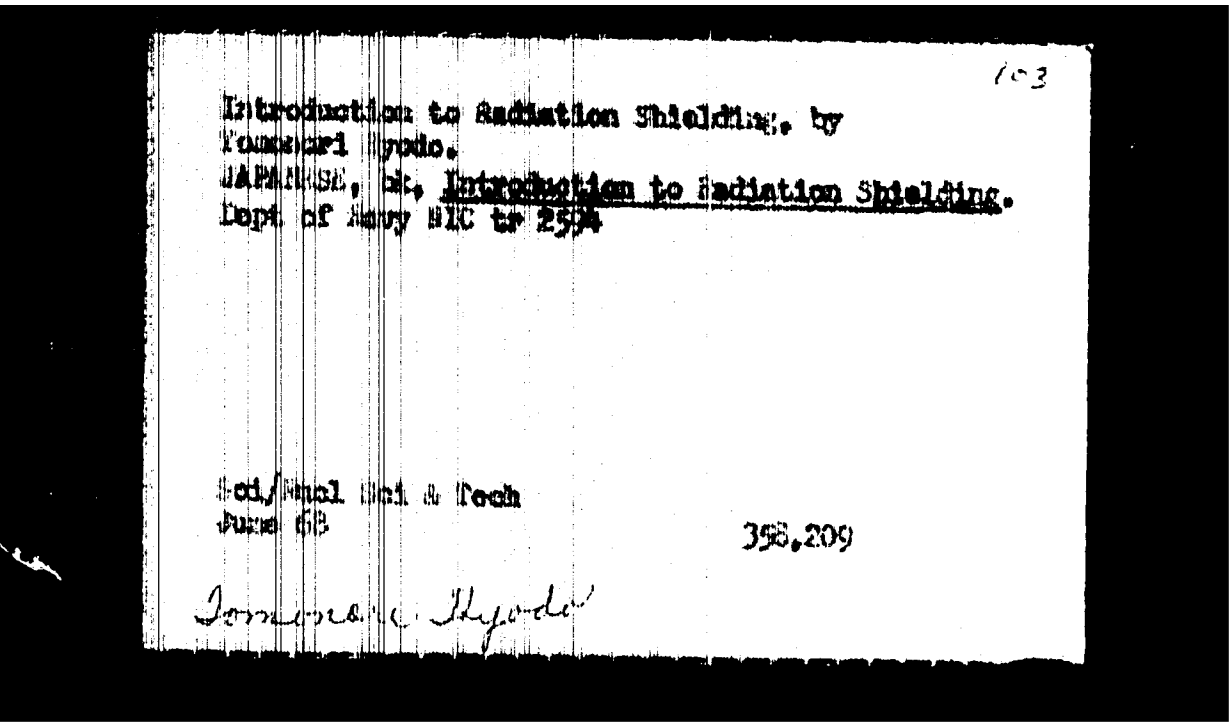

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Dept of Navy NIC tr 2594

Ed/Final Int & Tech
June 68

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Tsuneari Iyodo

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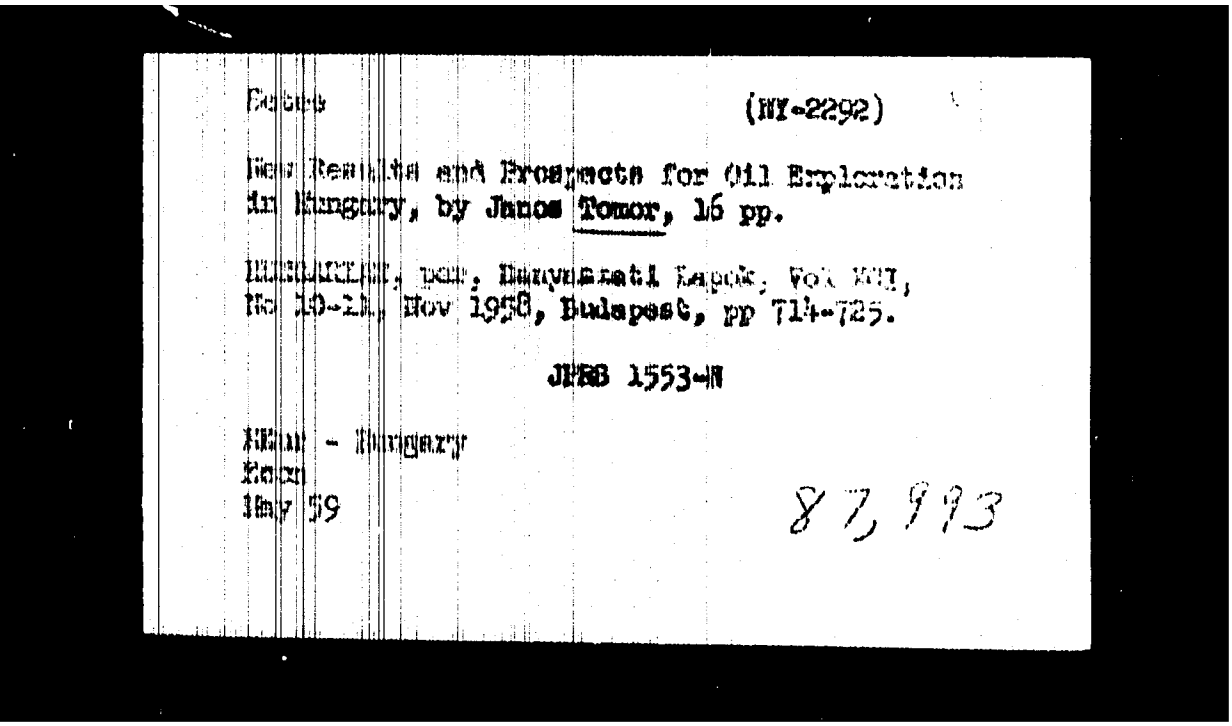

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Min - Hungary
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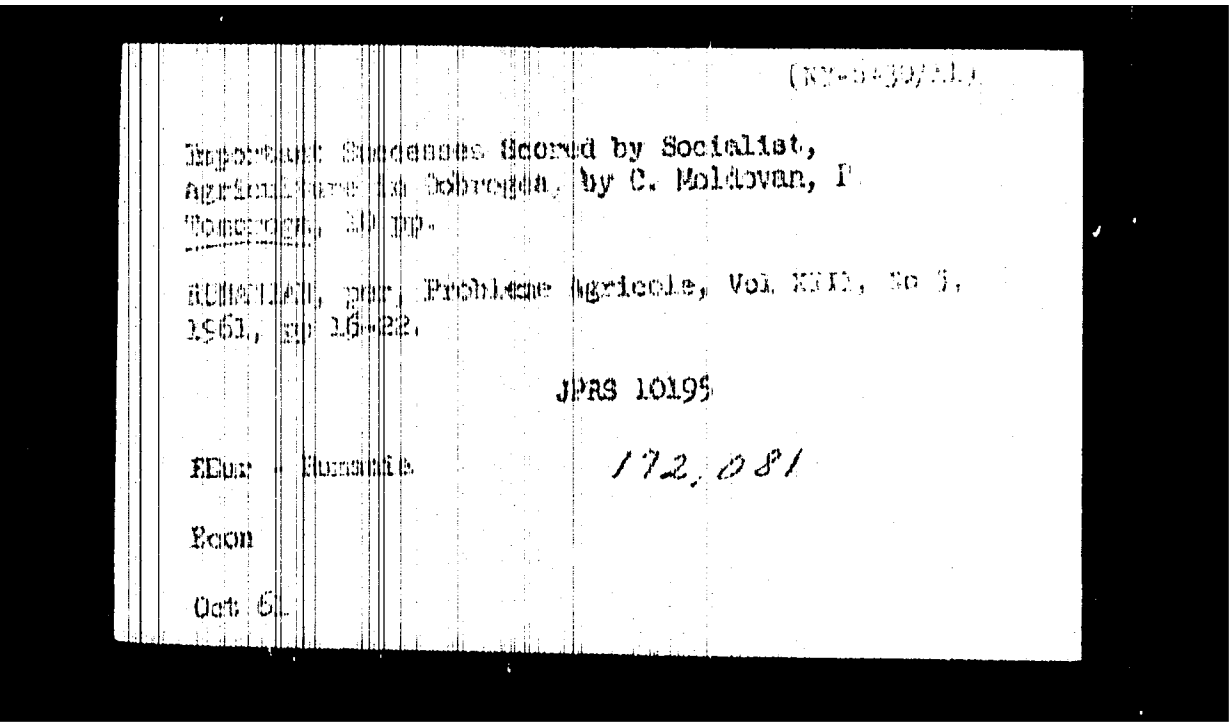

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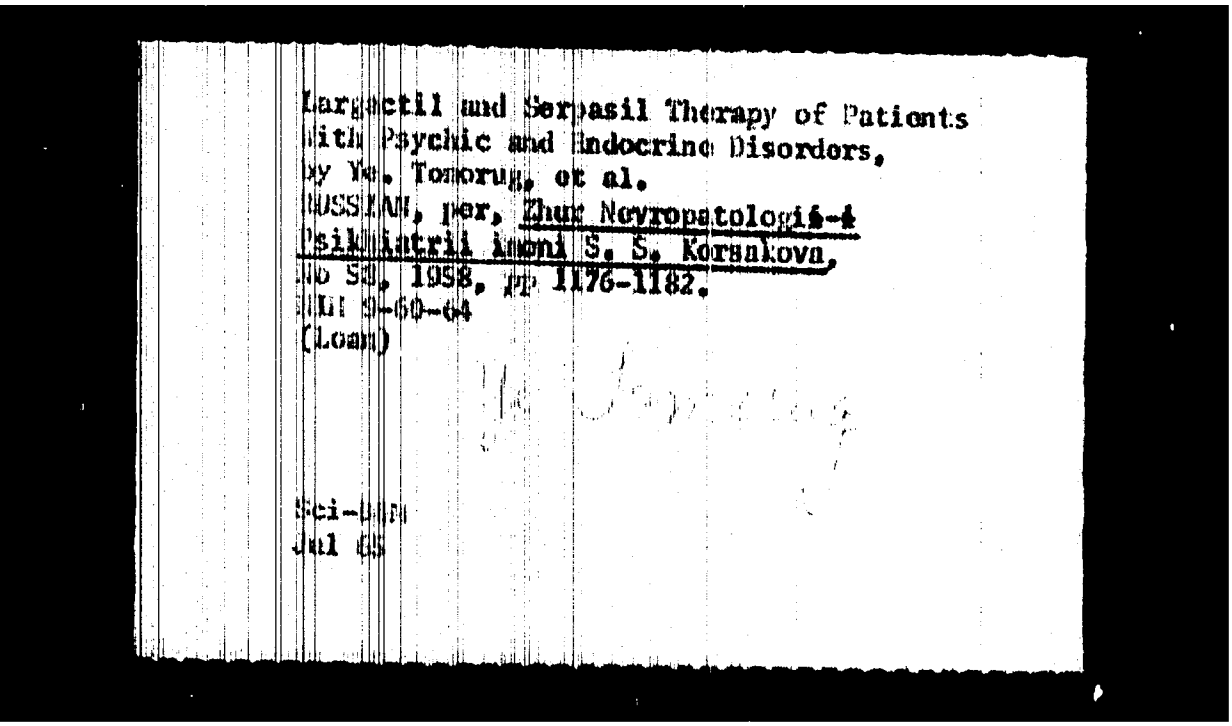

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MOSSCOW, per, Zhur Nevropatologii i
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No 58, 1958, pp 1176-1182.
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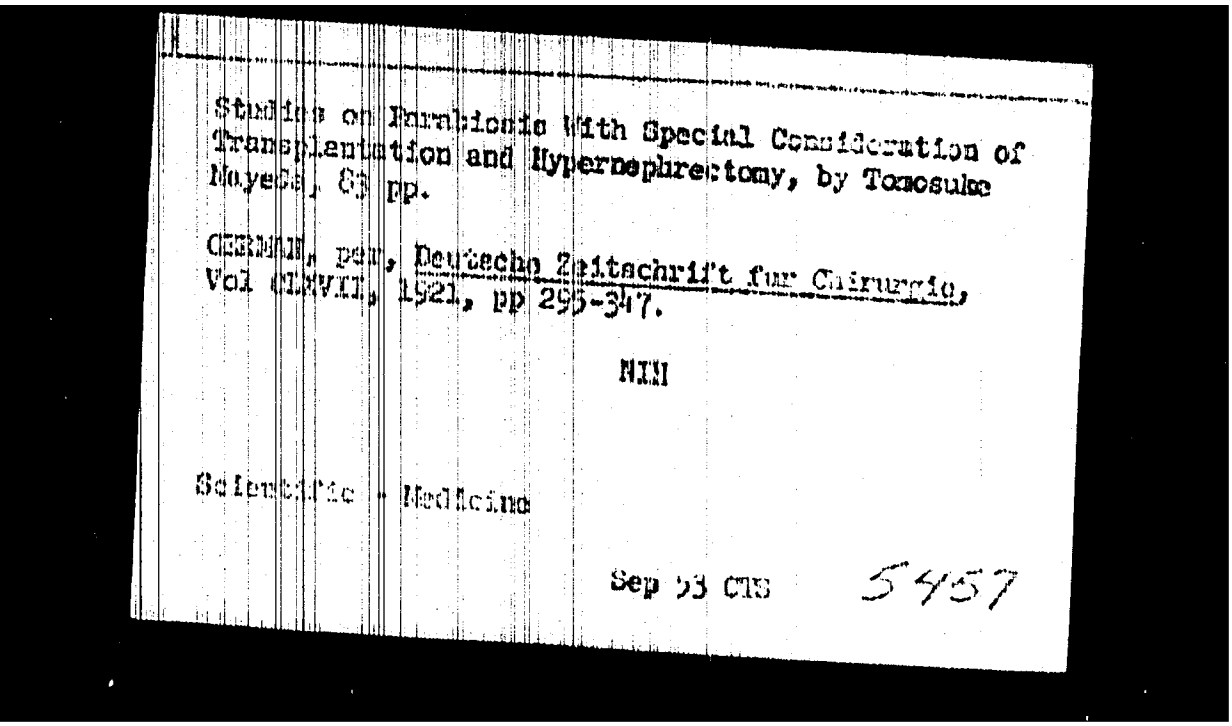

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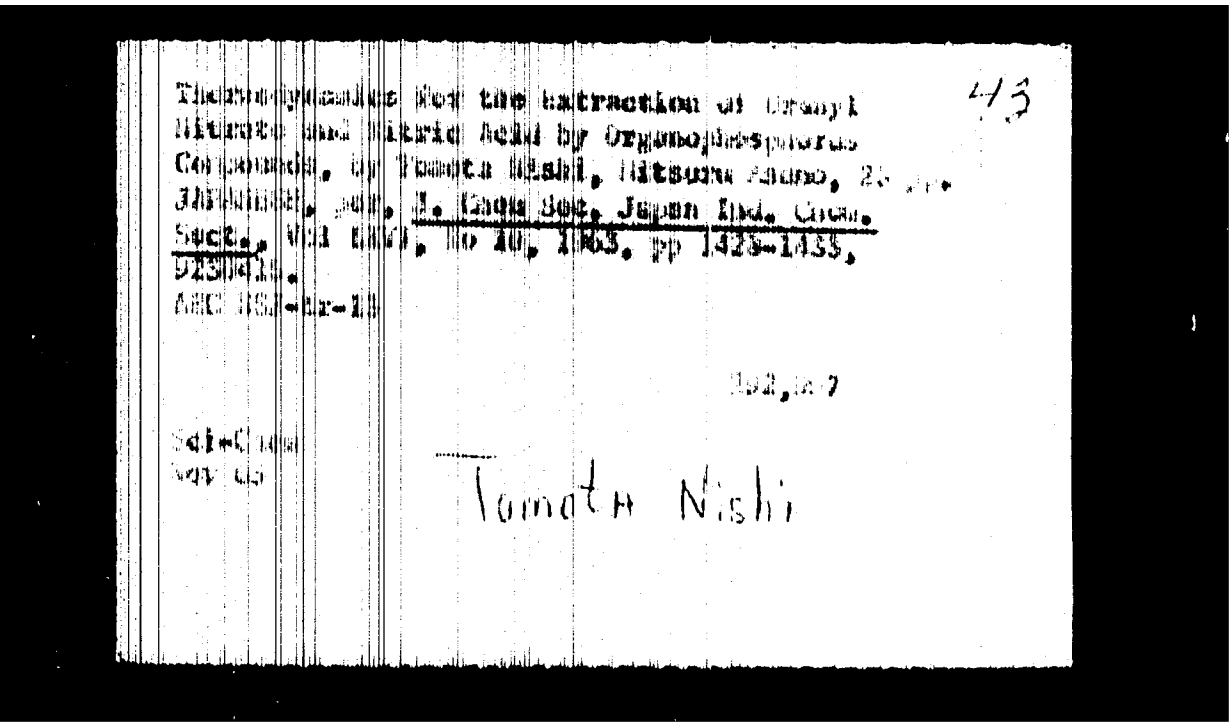

Tomoto Nishi

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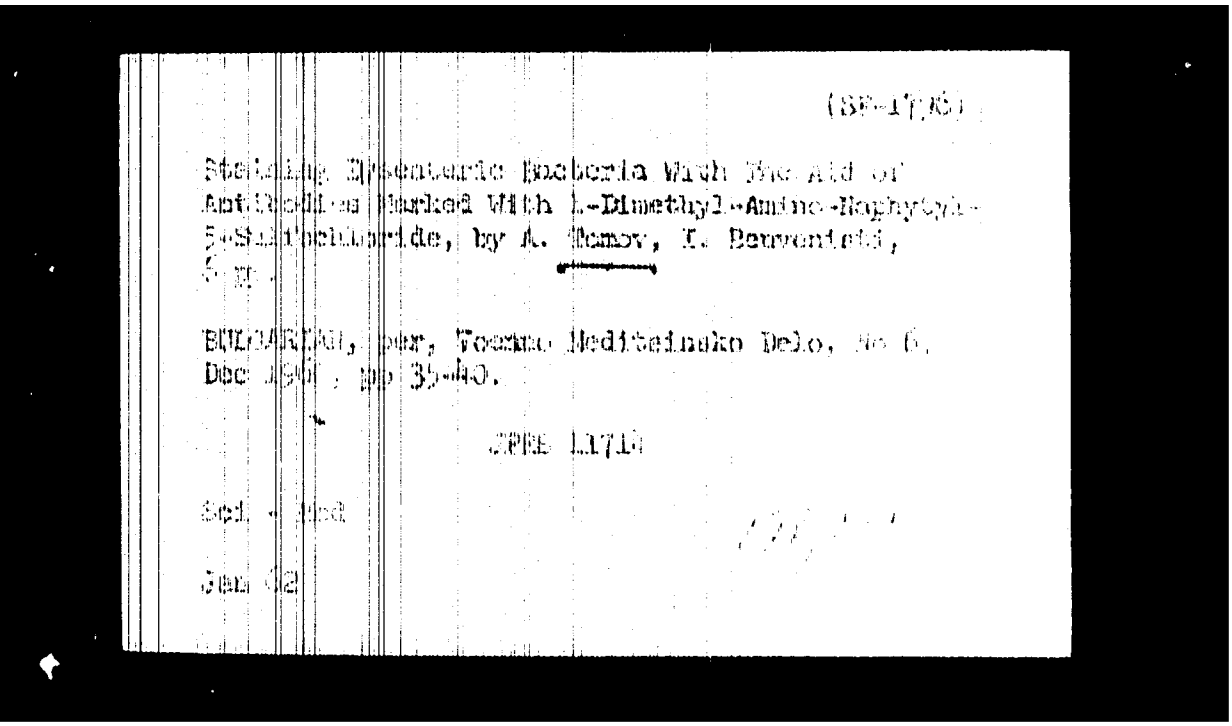

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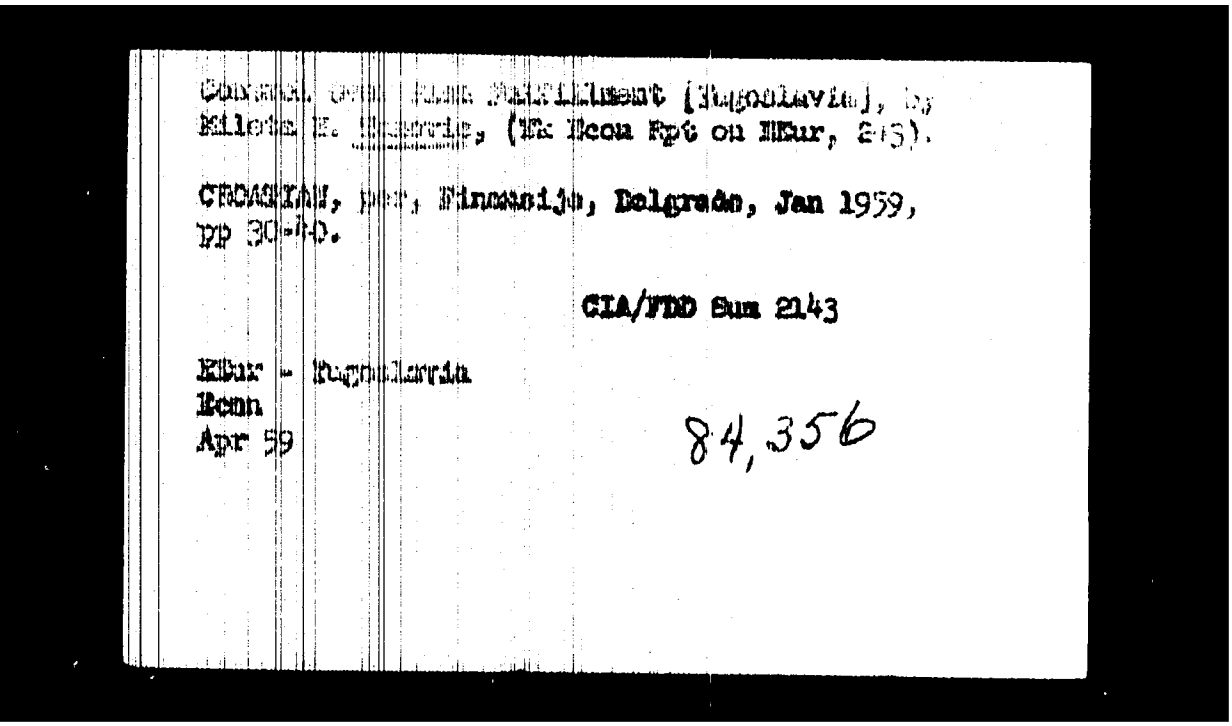

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EEur - Bulgaria
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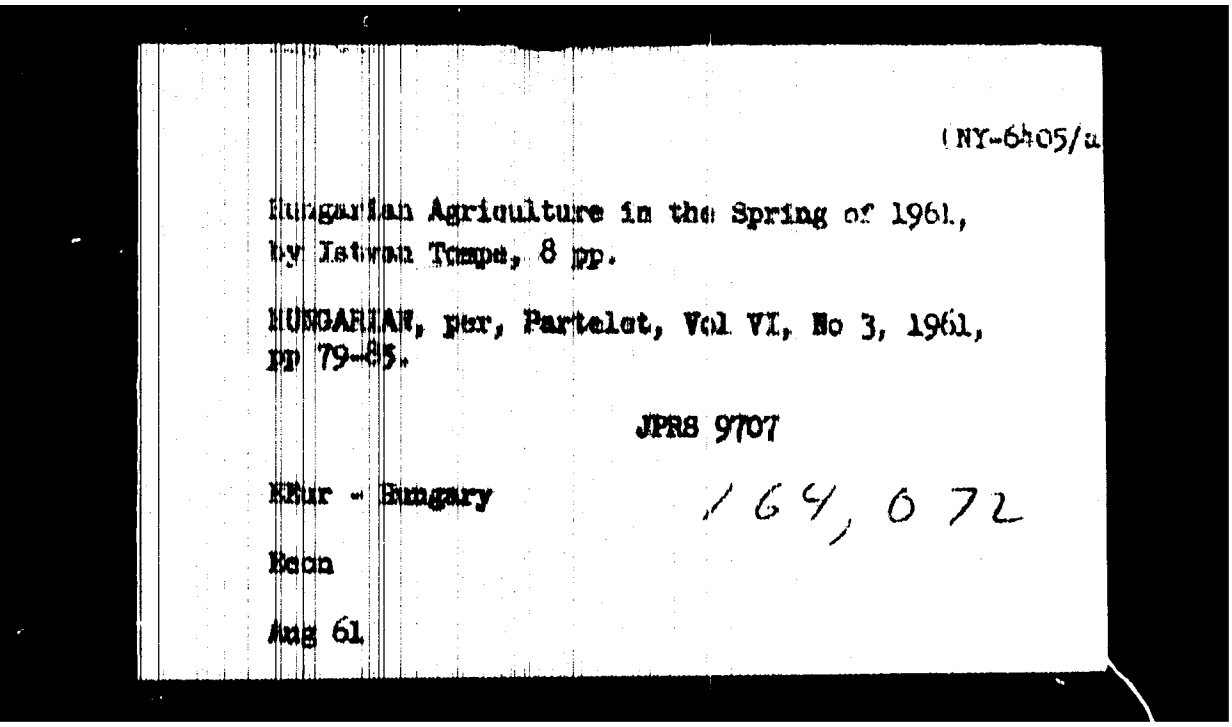

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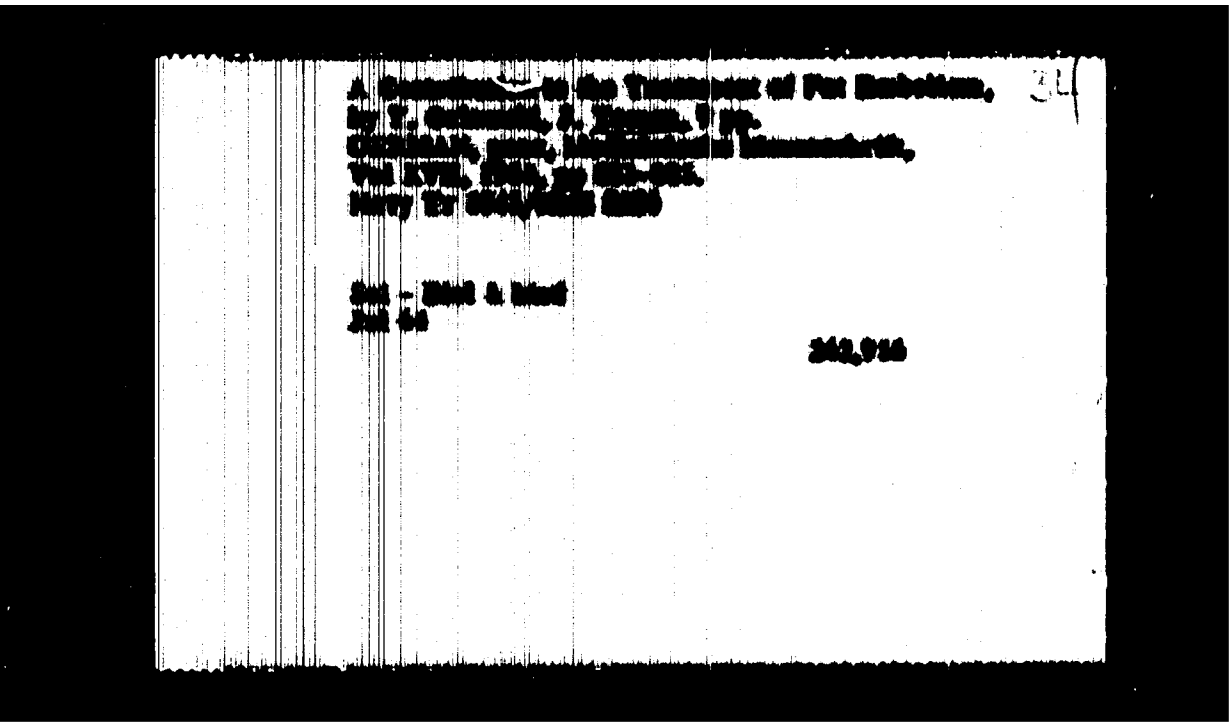

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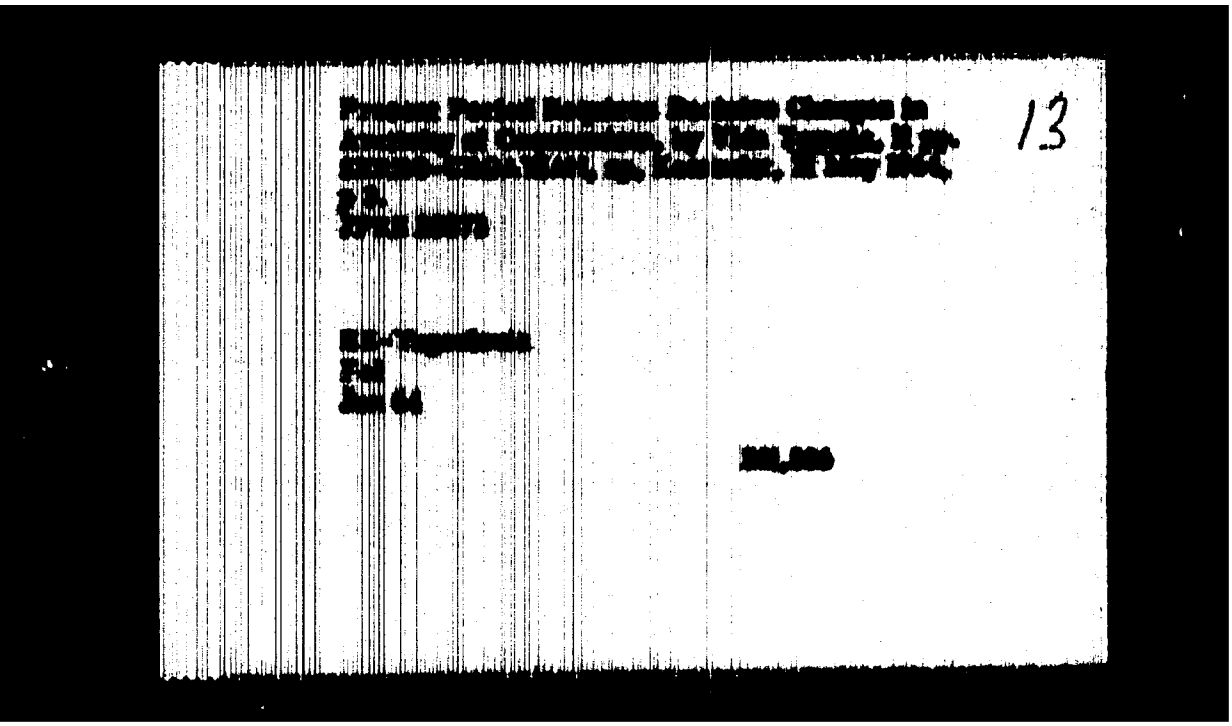

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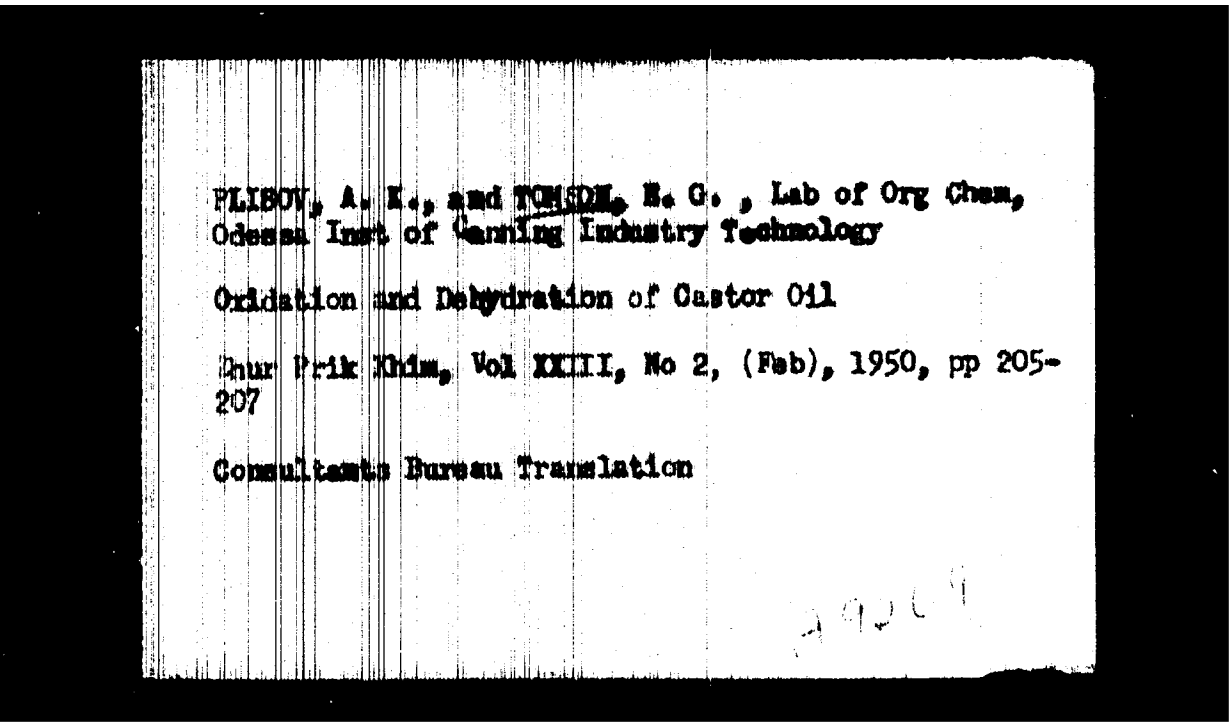

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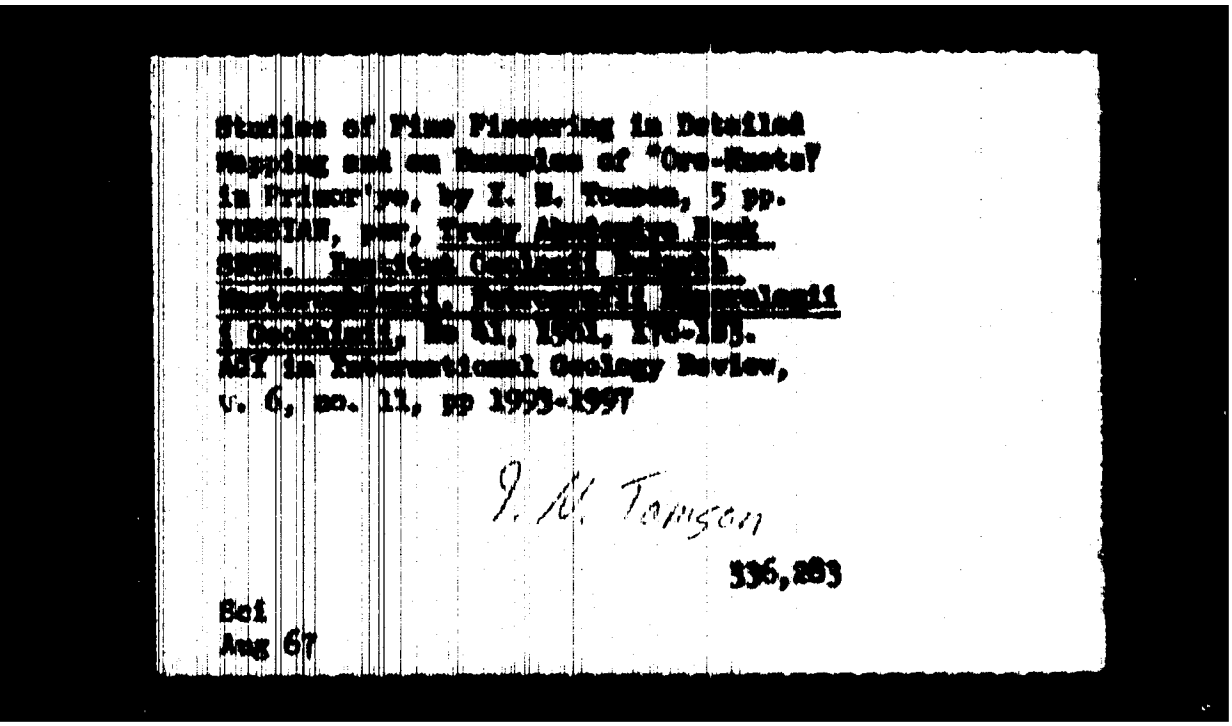

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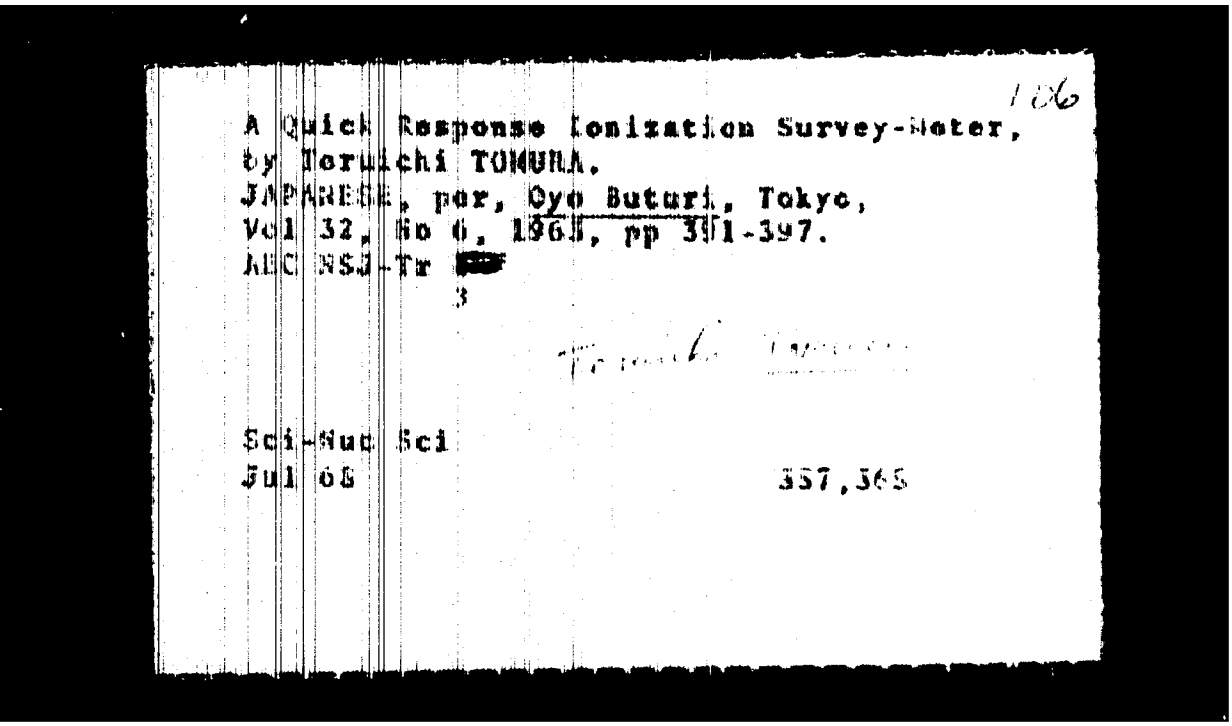

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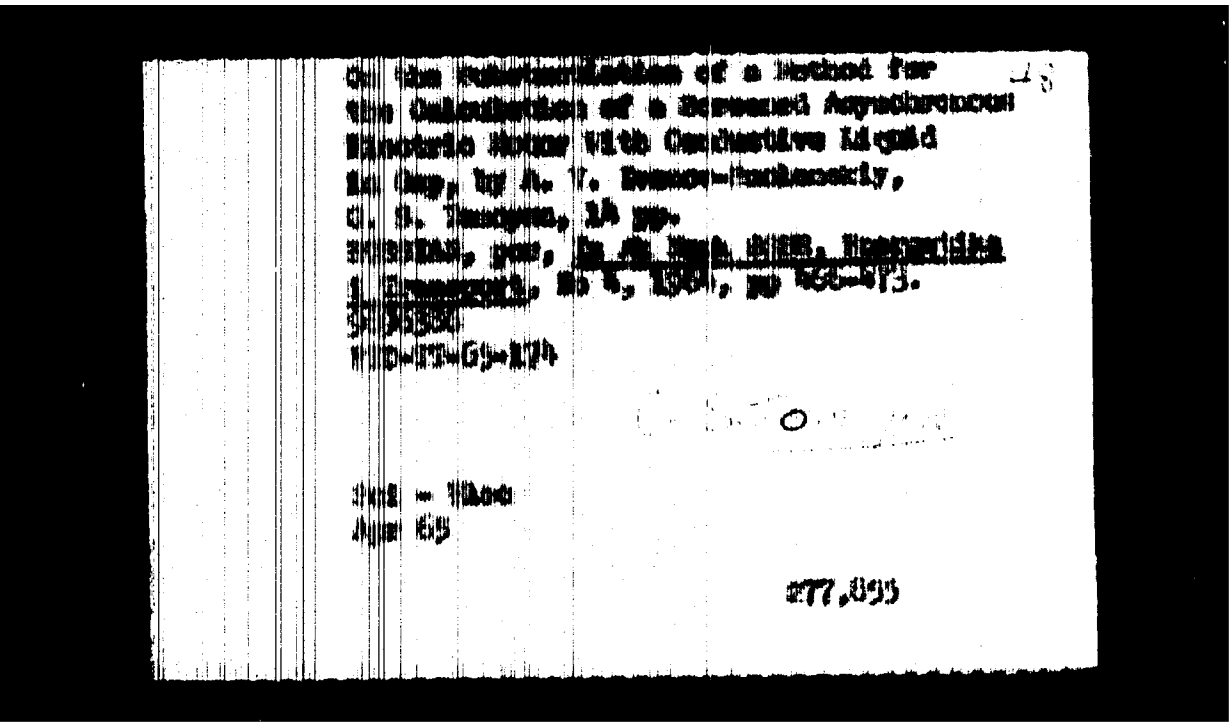

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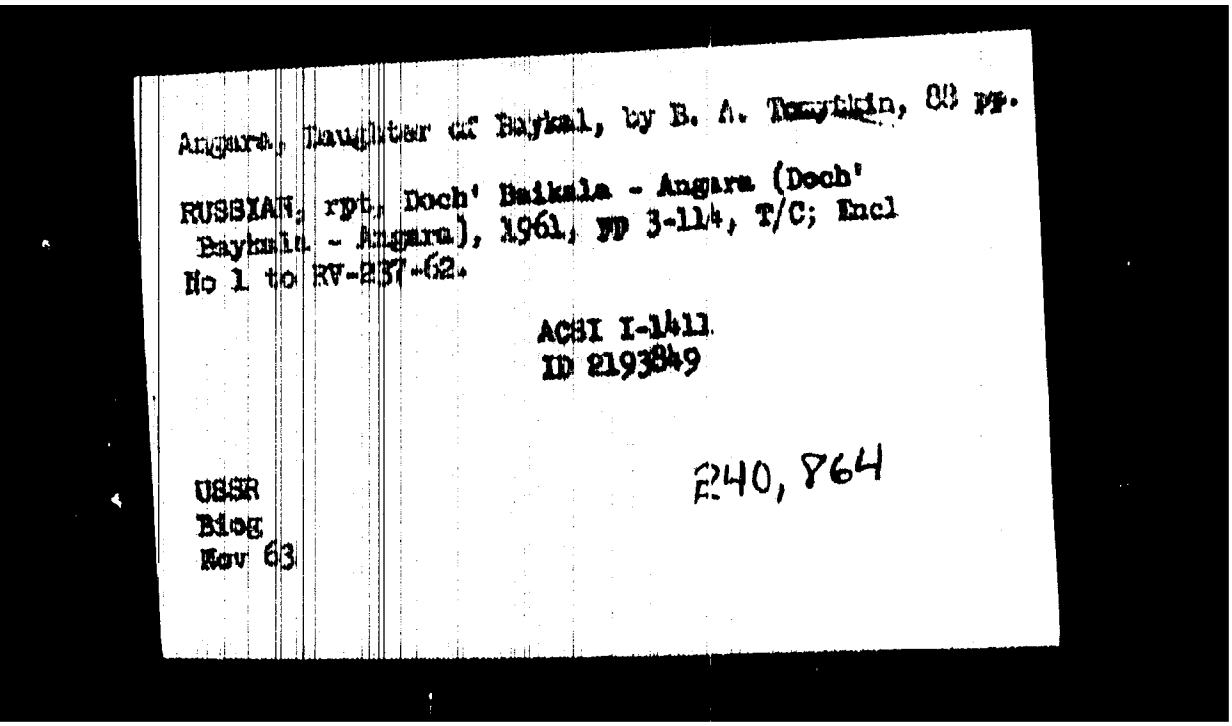

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ACSI I-1411
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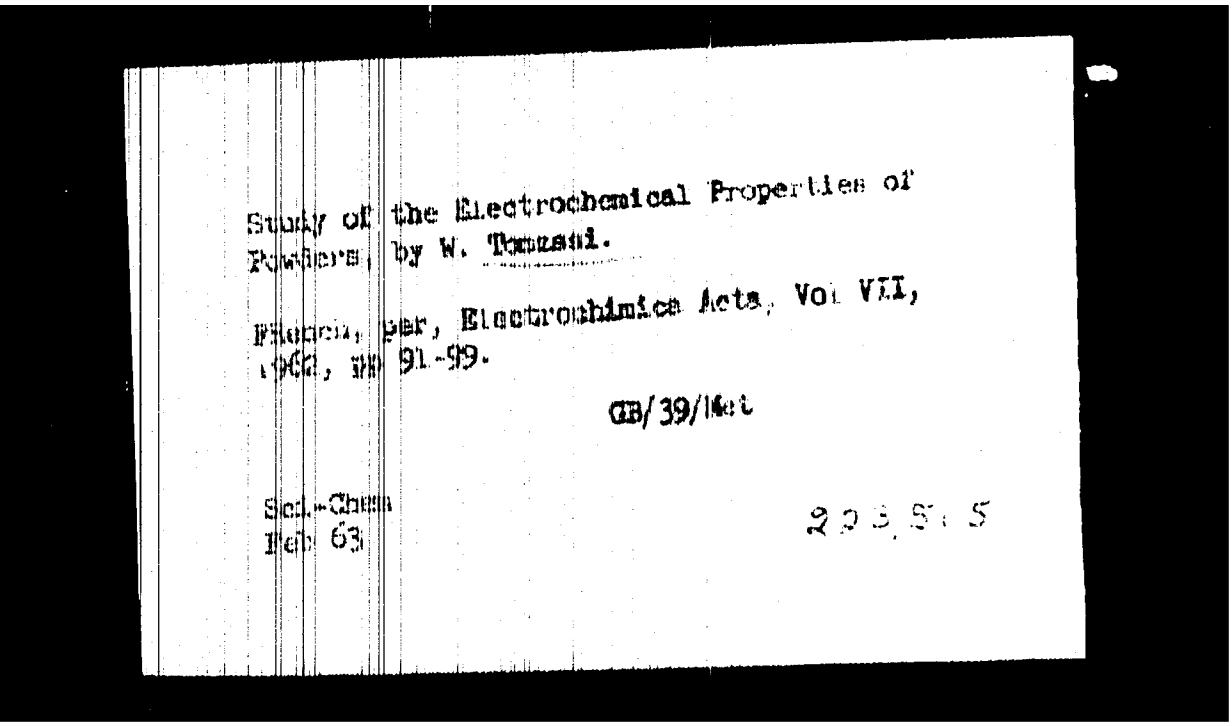

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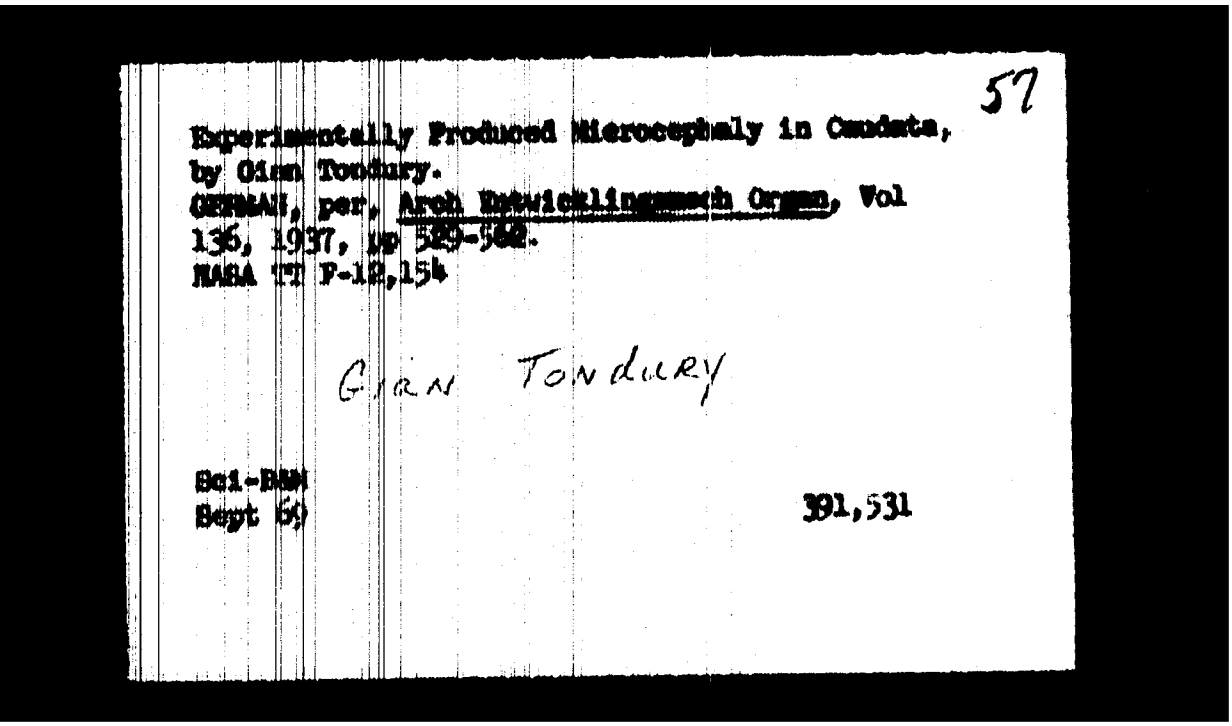

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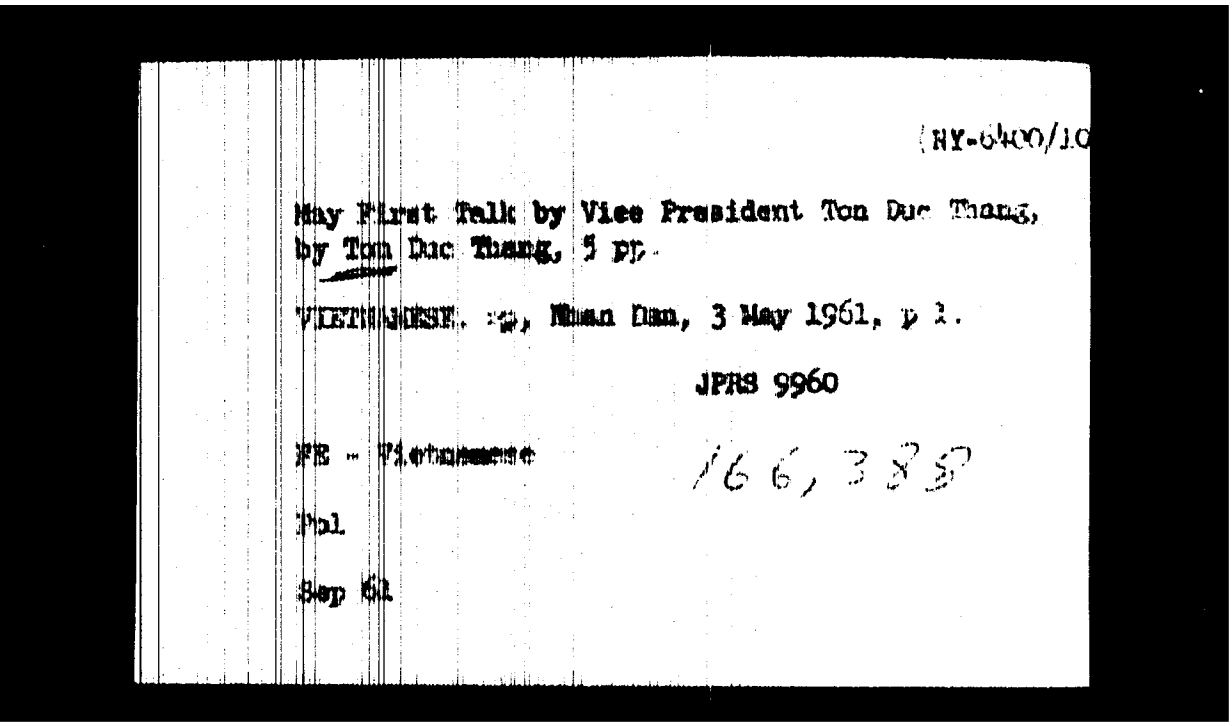

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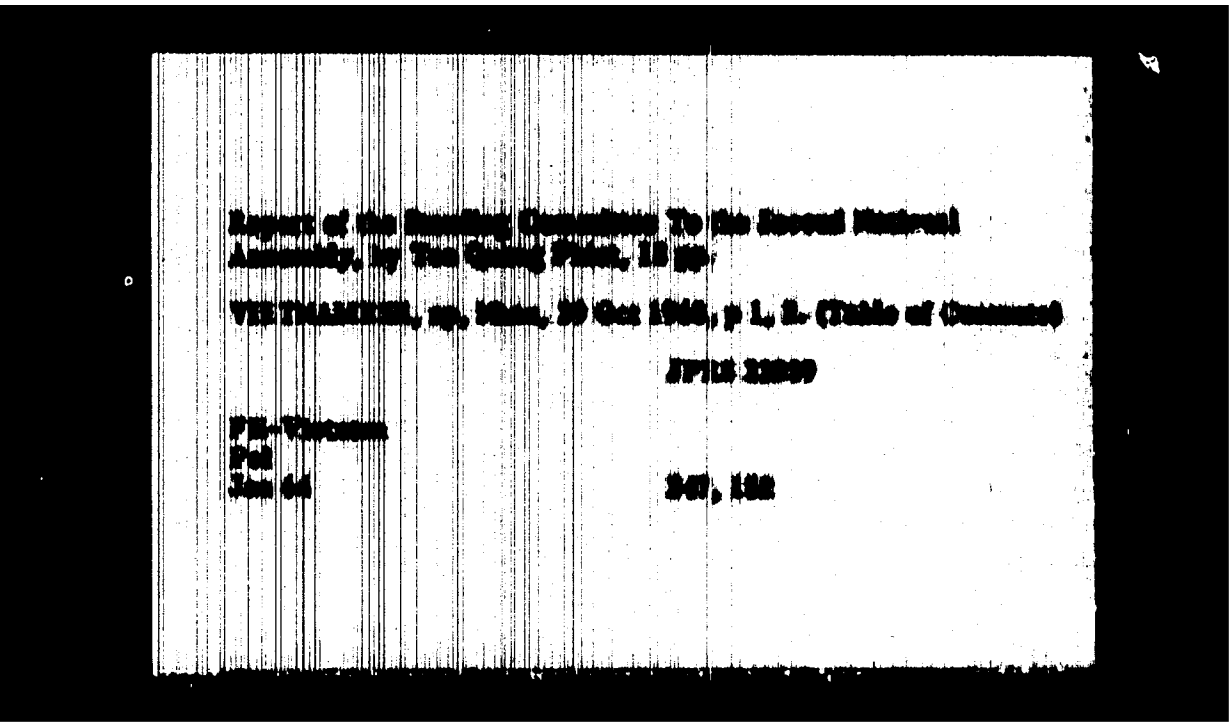

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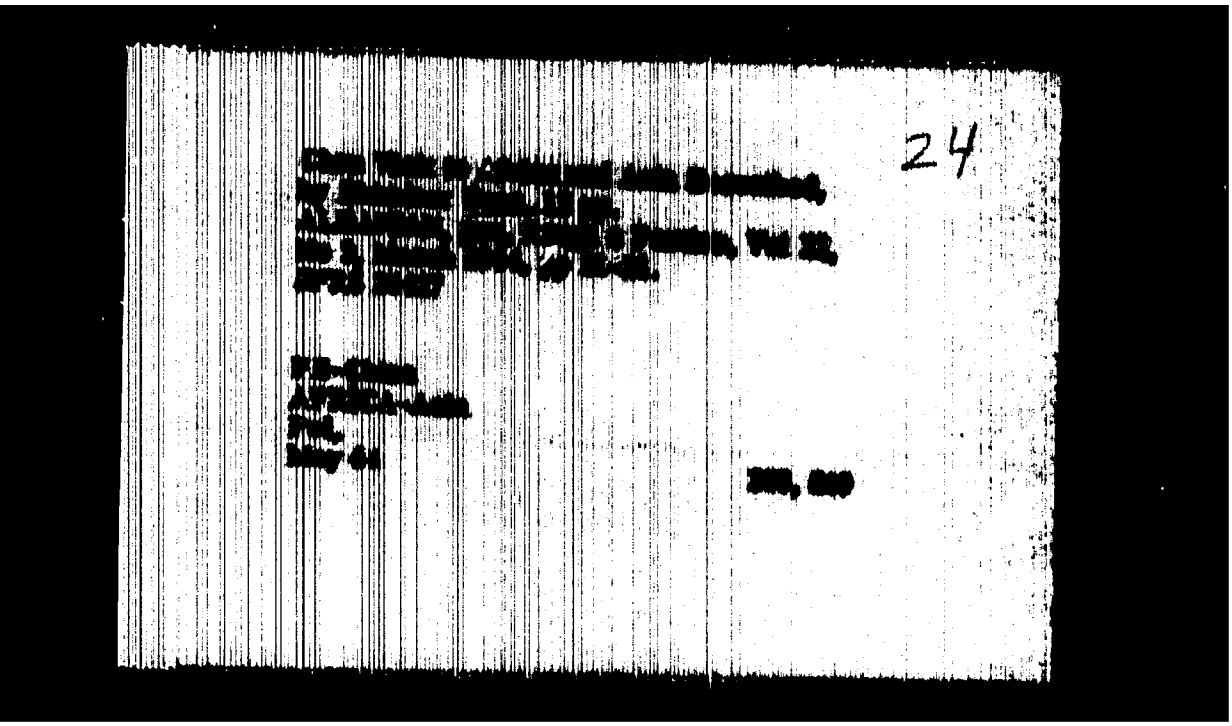

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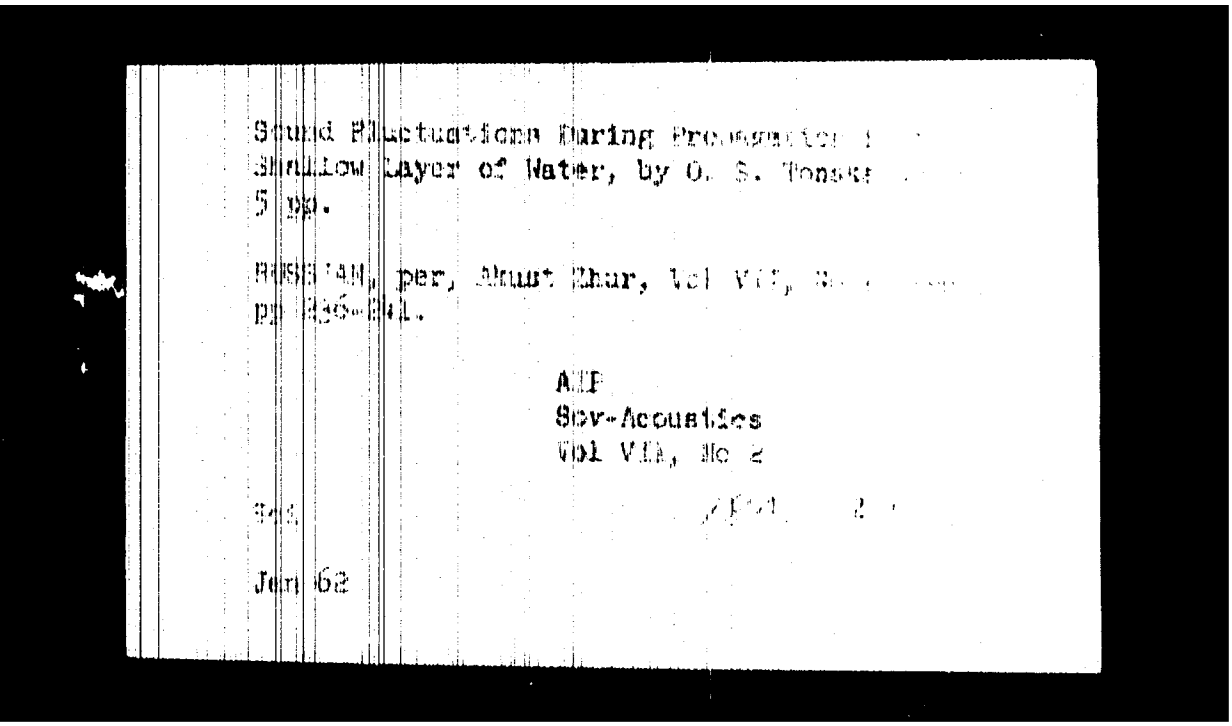

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A BEAD POLYMERIZATION PROCESS. July 63, 9p.
Order from SA $16.00 SA Code-P-112

Trans. of published (Japanese) patent 937/1962, cl. 26-B-011.3 (26-B-121) (26-B-151) appl. no. 21315/1960, filed 19 Apr 60, pub. 25 July 62, by Toyo Koatsu Industries, Inc.

DESCRIPTORS: *Styrene plastics, *Acrylic resins, *Polyvinyl alcohol, *Acrylonitriles, Polymerization, Particles, Production.

(Materials-Plastics, TT, v. 10, no. 6)

63-17943

I. Title: Bead polymerization
I. Tonami, N.
II. Ozaki, M.
III. Patent (Japan) pub. 37-0 337
IV. SA Code-P-112
V. Seizaburo Aoki (Japan)

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